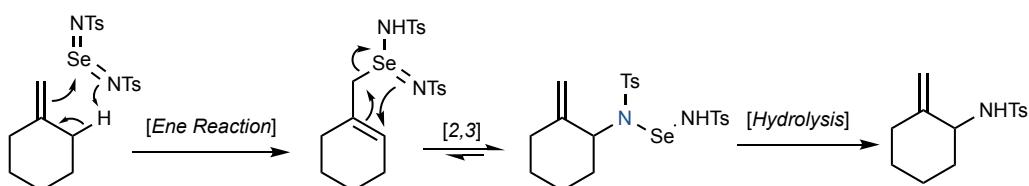
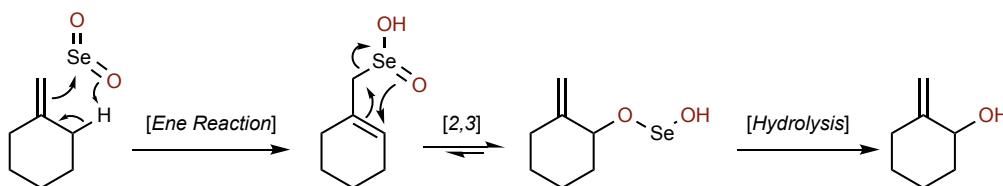
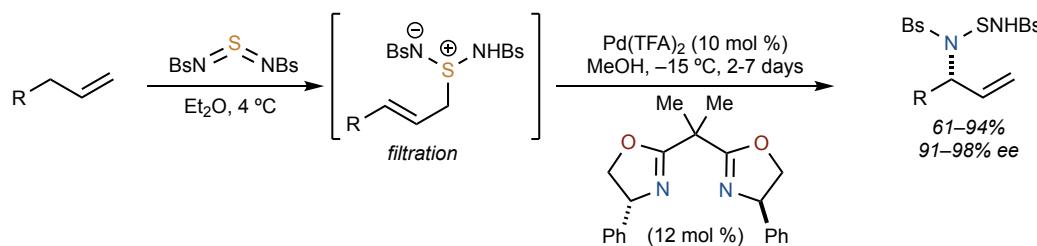


# Sharpless-Kresze Amination

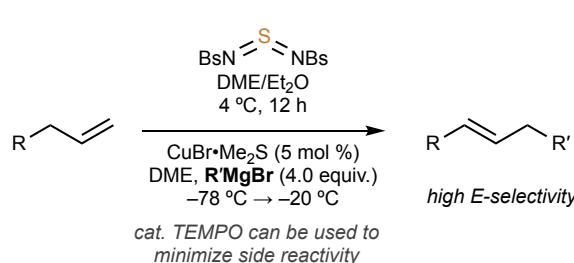


● much more reactive than  $\text{SeO}_2$    ● absence of rearranged products   ● positional selectivity similar to  $\text{SeO}_2$

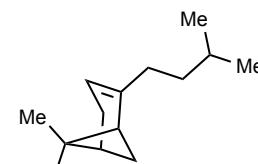
Tambar J. Am. Chem. Soc. 2012, 134, 18495; <https://doi.org/10.1021/ja307851b>



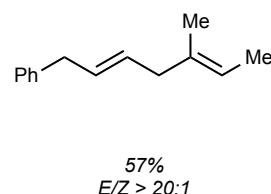
Tambar Angew. Chem. Int. Ed. 2014, 53, 1664; <https://doi.org/10.1002/anie.201309134>



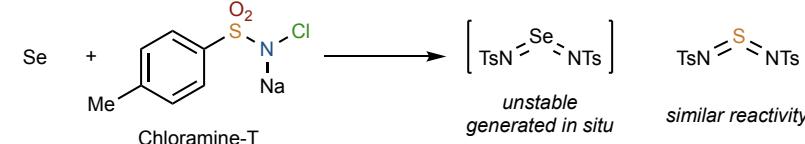
## Selected Examples



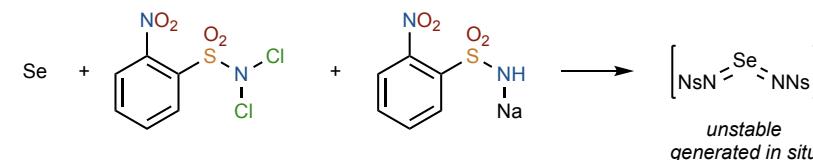
57%



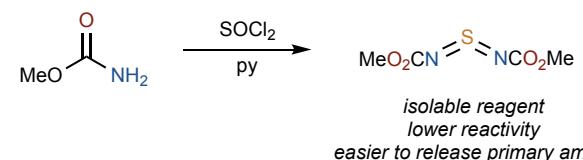
57%  
 $E/Z > 20:1$



Sharpless Angew. Chem. Int. Ed. 1996, 35, 454

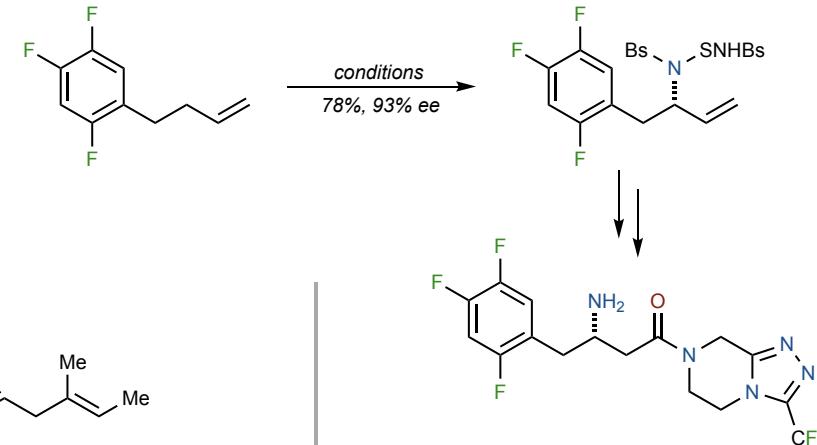


Kresze J. Org. Chem. 1983, 48, 3561



isolable reagent  
lower reactivity  
easier to release primary amine

Tambar Synlett 2013, 24, 2459



# Sharpless-Kresze Amination

Tambar *Nature* 2017, 547, 196; <https://doi.org/10.1038/nature22805>

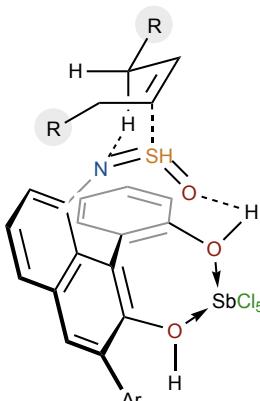
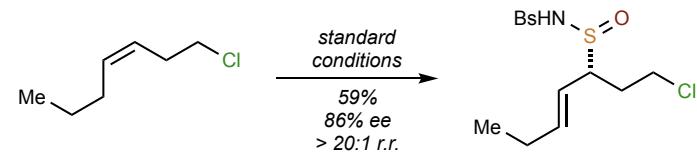
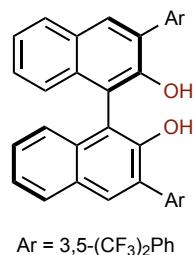
## Enantioselective Allylic Functionalization



Comparable FG: Ts-Indole, -OTFA, -Hal

Positional selectivity:  $\text{CH}_2 > \text{CH}_3 > \text{CH}$

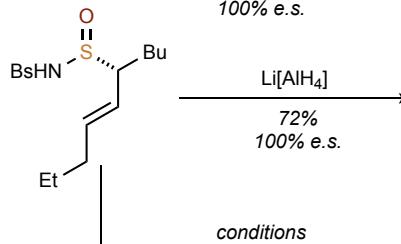
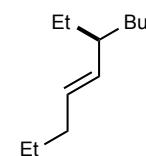
Positional selectivity: sensitive to electronic effects



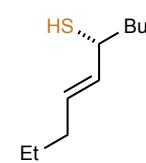
- Lewis-acid-assisted Bronsted acidity (LBA)
- Protonation lowers LUMO
- Cis-olefin provides higher enantioselectivity
- Exo transition state

$\text{CuBr}\cdot\text{Me}_2\text{S}$  (5 mol %)

EtMgCl  
73%  
3.1 r.r.  
100% e.s.



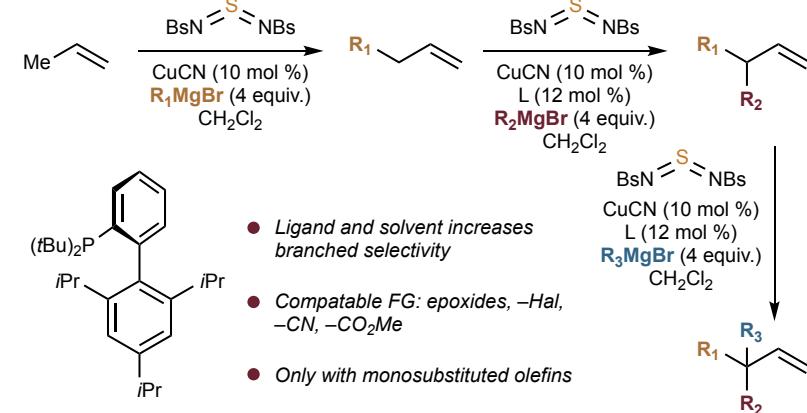
conditions



$X = \text{OH}$        $\text{Me}_2\text{SO}_4$ ,  $\text{Et}_3\text{N}$ ;  
                 $\text{PhMgBr}$ ;  $\text{P}(\text{OEt})_3$ , 74%;  
 $X = \text{NHBS}$        $\text{TiCl}(\text{O}-\text{i-Pr})_3$ ;  $\text{P}(\text{OEt})_3$ , 71%;  
 $X = \text{Cl}$              $\text{SO}_2\text{Cl}_2$ , 94%, 3:1 r.r.

Tambar *J. Am. Chem. Soc.* 2019, 141, 17305; <https://doi.org/10.1021/jacs.9b08801>

## Sequential Allylic Alkylation



## Enantioselective Allylic Alkylation

