

## Outline:

1. Introduction
2. Synthesis of Normal NHCs
3. General Reactivity
  - 3.1 Breslow intermediate
  - 3.2 Acyl Enolate
  - 3.3 Acyl Azolium
  - 3.4 Single Electron Transfer
  - 3.5 Miscellaneous Activation mode
4. Total Synthesis Examples with NHCs

## Not included

- NHC as Metal Ligand
- NHC as C1 Synthons
- Brønsted Base Catalysts

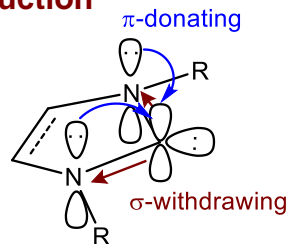
## Relevant Reviews

- Enders, D. et al. *Chem. Rev.* 2007, 107, 5606–5655. <https://doi.org/10.1021/cr068372z>
- César, V. et al. *Chem. Rev.* 2011, 111, 2705–2733. <https://doi.org/10.1021/cr100328e>
- Glorius, F. et al. *Nature.* 2014, 485–496. <https://doi.org/10.1038/nature13384>
- Rovis, T. et al. *Chem. Rev.* 2015, 115, 9307–9387. <https://doi.org/10.1021/acs.chemrev.5b00060>
- Chen, X. et al. *Chin. J. Chem.* 2020, 38, 1167–1202. <https://doi.org/10.1002/cjoc.202000107>

## Book

Buju, A. T., *N-Heterocyclic Carbene in Organocatalysis*. Wiley, 2018.

## Introduction

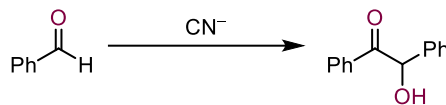


### Heterocycle Back bone

- Aromaticity (nonessential) ~ 25 kcal/mol stabilization
- Change the electronic property

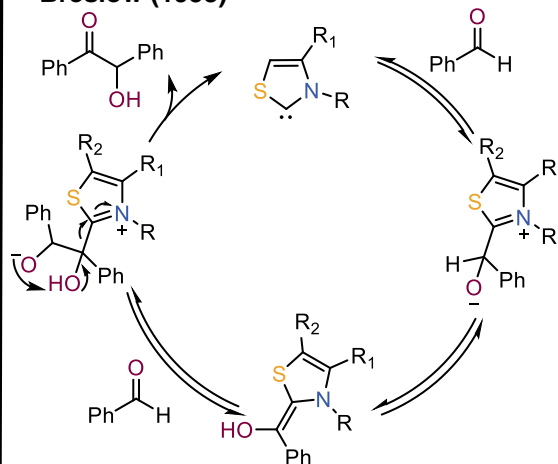
Glorius, F. et al. *Nature.* 2014, 485–496. <https://doi.org/10.1038/nature13384>

### Wöhler, Liebig (1832)



Wöhler, F.; Liebig, J. *Ann. Der. Pharm.* 1832, 3, 249–287.

### Breslow (1958)



### Breslow intermediate

Breslow, R. *J. Am. Chem. Soc.* 1958, 80, 3719–3726. (1958). <https://doi.org/10.1021/ja01547a064>

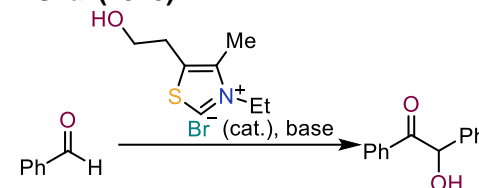
### N-Substituted

- Prevent of dimerization
- Change the electronic property
- Asymmetric induction

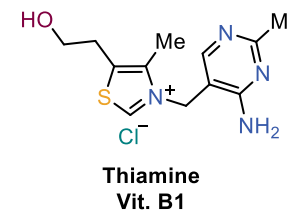
### Ring Size

- Change the Bent angle of carbene
- Cyclic structure favor singlet carbene

### Ukai (1943)

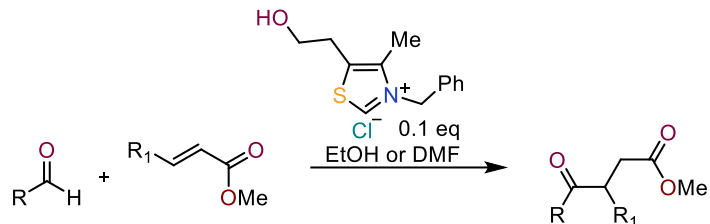


Ukai, T. et al. *I. J. Pharm. Soc. Jpn.* 1943, 63, 296–304



- Coenzyme of pyruvate decarboxylase
- Converting pyruvate to acetoin
- Catalyzing transketolase

## Stetter (1976)



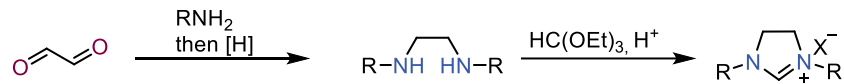
R, R<sub>1</sub>=Alkyl, Ar

Stetter, H. et al. *Angew. Chem. Int. Ed.* 1973, 12, 81. <https://doi.org/10.1002/anie.197300811>

## Synthesis of Normal NHCs

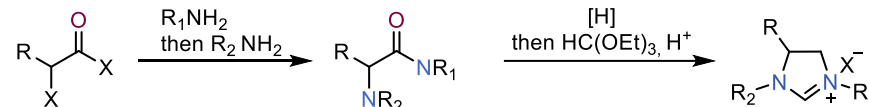
### Imidazolinylidene

symmetrical



Arduengo III, A. J. *Tetrahedron* 1999, 55, 14523. [https://doi.org/10.1016/S0040-4020\(99\)00927-8](https://doi.org/10.1016/S0040-4020(99)00927-8)

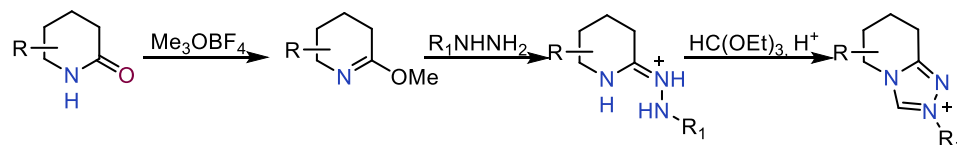
unsymmetrical



$\alpha$ -halogenoacyl core

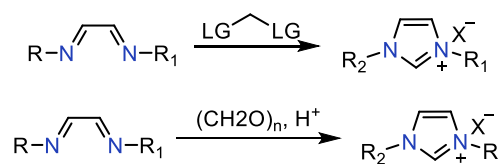
Kotschy, A. et al. *J. Org. Chem.* 2006, 71, 5969-597. <https://doi.org/10.1021/jo060594+>

### Triazolium



Leeper, F. J. et al. *J. Chem. Soc., Perkin Trans. 1*, 1998, 1891-1894. <https://doi.org/10.1039/A803635G>

### Imidazolium

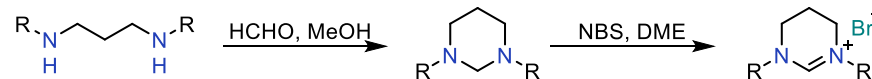


**common C1-synthons**

CH<sub>2</sub>I<sub>2</sub>  
[Ph<sub>3</sub>As-CH<sub>2</sub>-OTf]<sup>+</sup>(TfO)<sup>-</sup>  
Cl-CH<sub>2</sub>-OMe  
tBuCO<sub>2</sub>-CH<sub>2</sub>-OTf

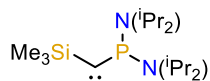
César, V. et al. *Chem. Rev.* 2011, 111, 2705-2733. <https://doi.org/10.1021/cr100328e>

### Tetrahydropyrimidinium



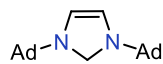
Buchmeiser, M. R. et al. *Macromol. Rapid Commun.* 2004, 25, 231-236. <https://doi.org/10.1002/marc.200300173>

## First Isolable Stabilized Carbene



Bertrand (1988)

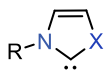
Bertrand, G. *J. Am. Chem. Soc.* 1988, 110, 6463-6466. <https://doi.org/10.1021/ja00227a028>



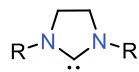
Arduengo (1991)

Arduengo III, A. J. *J. Am. Chem. Soc.* 1991, 113, 1, 361-363. <https://doi.org/10.1021/ja00001a054>

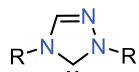
## Common NHC Class



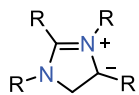
X=NR, Imidazolylidene  
X=S, Thiazolylidene  
X=O, Oxazolylidene



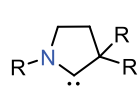
Imidazolinylidene



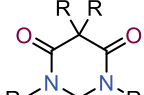
Triazolylidene



Abnormal  
imidazolylidene



Cyclic alkyl amino  
carbene (CACC)



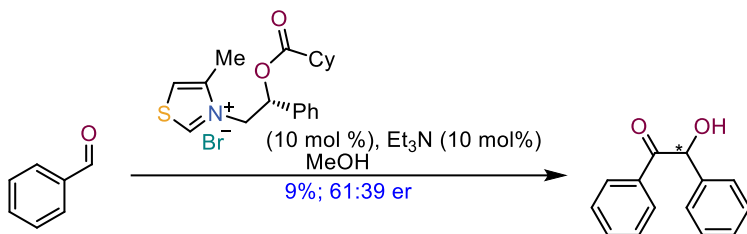
N,N'-diaminocarbene  
(DAC)

Glorius, F. et al. *Nature*. 2014, 485-496. <https://doi.org/10.1038/nature13384>

## General Reactivity

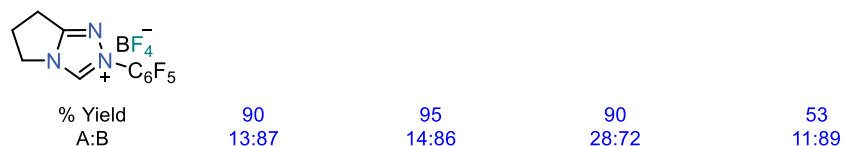
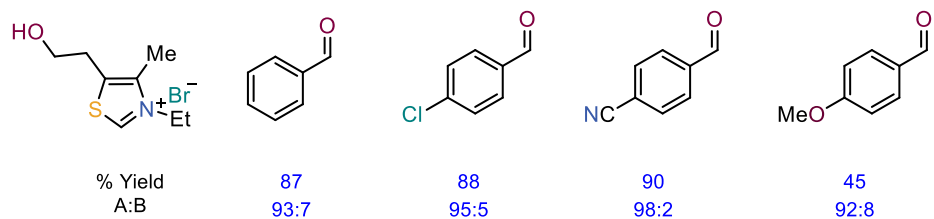
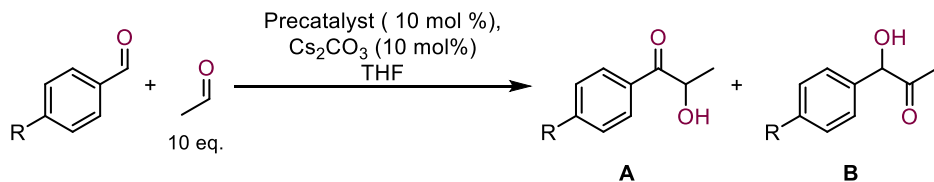
### Breslow intermediate

### First Example of Asymmetric Benzoin Condensation



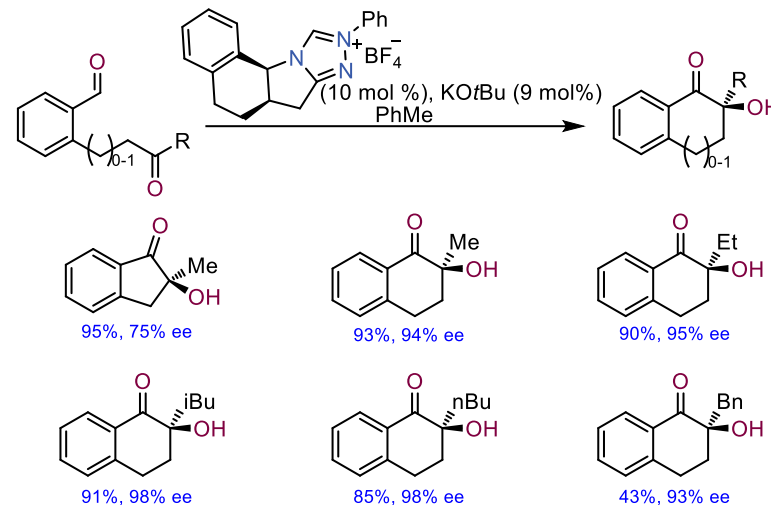
Sheehan, J. C. *J. Am. Chem. Soc.* 1966, 88, 15, 3666–3667. <https://doi.org/10.1021/ja00967a049>

### Different Catalyst Override Innate Reactivity



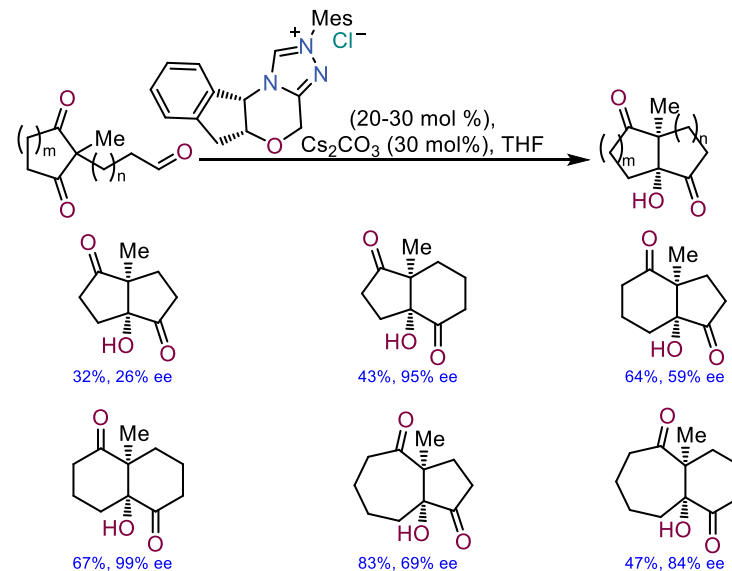
Yang, J.W.. Et al. *Org. Lett.* 2011, 13, 880–883. <https://doi.org/10.1021/ol102937w>

### Intramolecular Benzoin Condensation



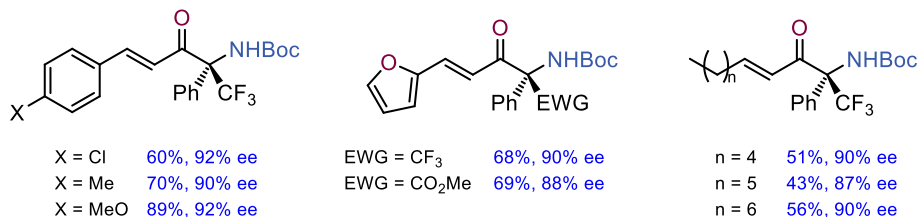
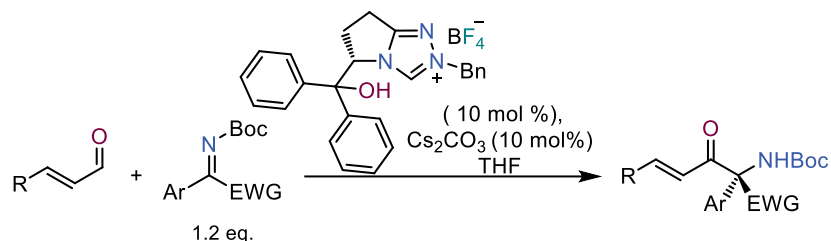
Enders, D. et al. *Angew. Chem. Int. Ed.* 2006, 45, 1463–1467. <https://doi.org/10.1002/anie.200503885>

### Desymmetrization of 1,3-Diketone by Benzoin condensation



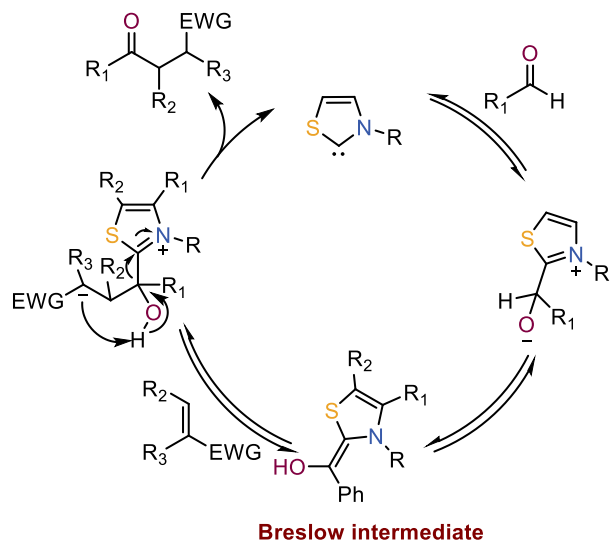
Ema, T. et al. *Org. Lett.* 2009, 11, 4866–4869. <https://doi.org/10.1021/ol901929z>

## Aza-Benzoin Condensation



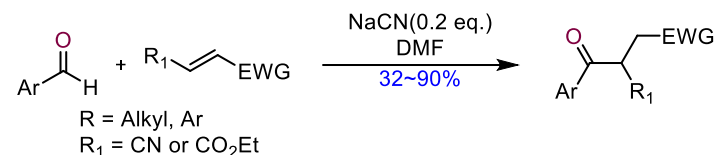
Ye, S. et al. *Angew. Chem. Int. Ed.* 2013, 52, 5803–5806. <https://doi.org/10.1002/anie.201301304>

## Stetter reaction Mechanism



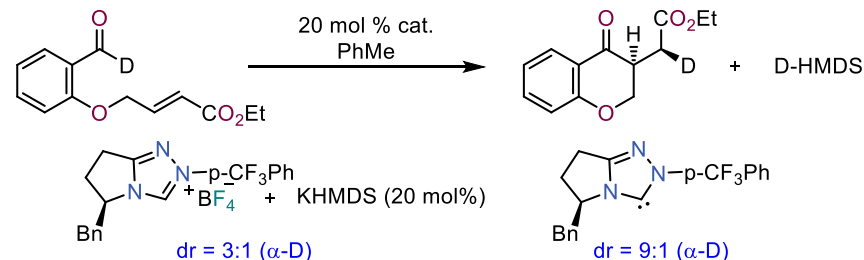
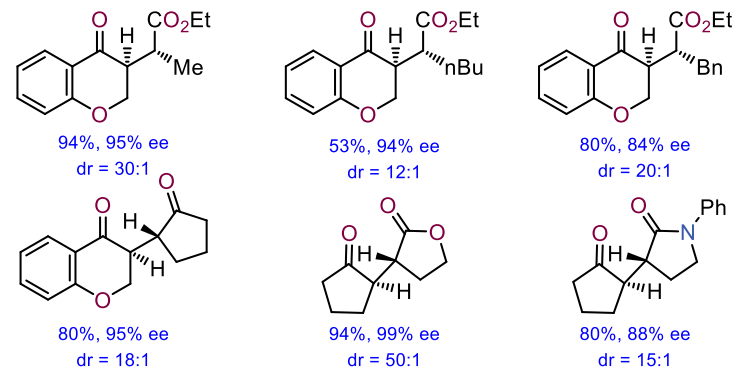
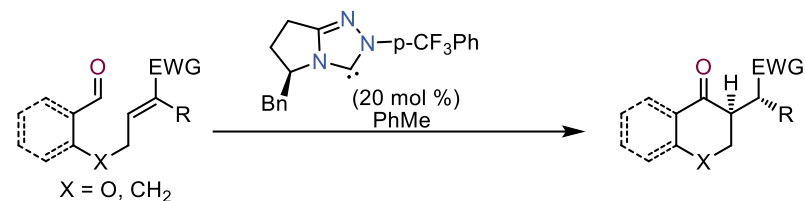
Rovis, T. et al. *Org. Lett.* 2011, 13, 7, 1742–1745. <https://doi.org/10.1021/ol200256a>

## First Report of Stetter Reaction



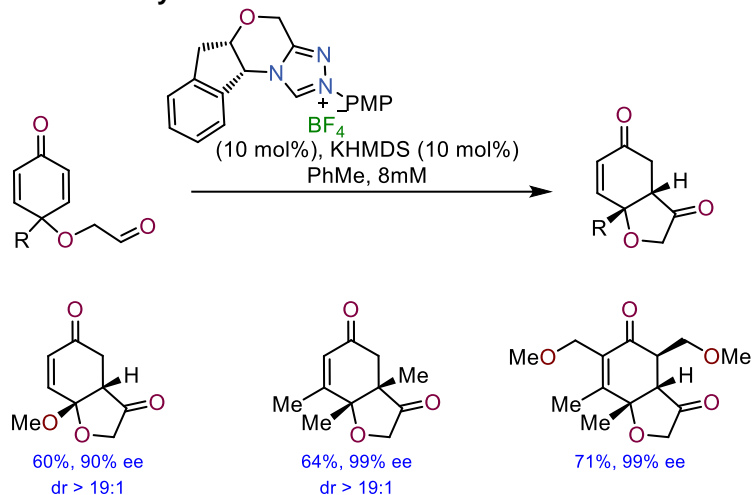
Stetter, H. et al. *Angew. Chem. Int. Ed.* 1973, 12, 81. <https://doi.org/10.1002/anie.197300811>

## Intramolecular Stetter Reaction



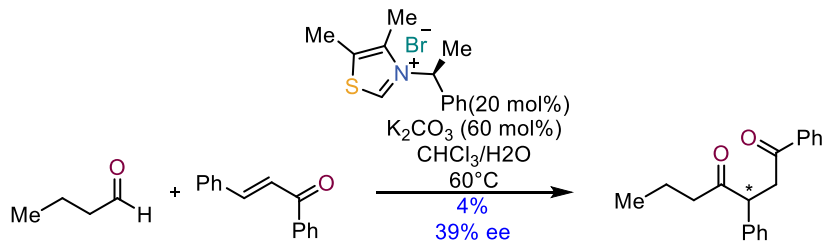
Rovis, T. et al. *J. Am. Chem. Soc.* 2005, 127, 17, 6284–6289. <https://doi.org/10.1021/ja042513z>

## Desymmetrization by Stetter Reaction

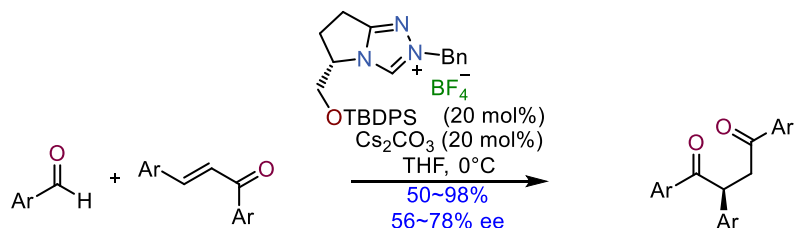


Rovis, T. et al. *J. Am. Chem. Soc.* 2006, 128, 8, 2552–2553. <https://doi.org/10.1021/ja058337u>

## Intermolecular Stetter reaction

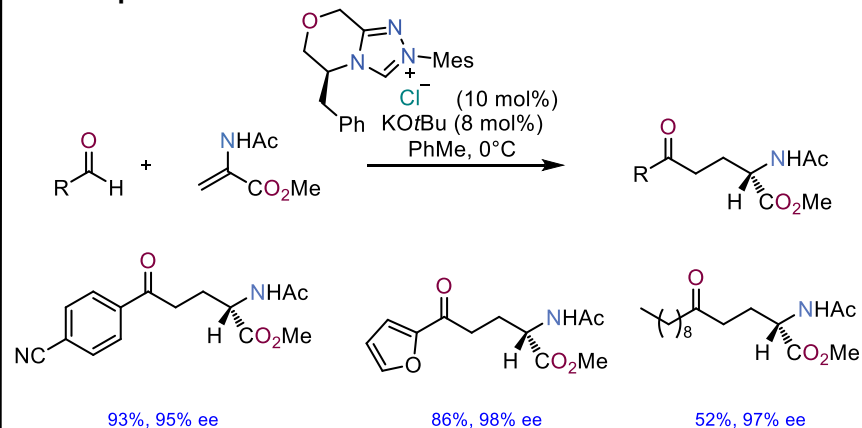


Enders, D. (1993). *Stereoselective Synthesis*, 63–90. Heidelberg, Germany: Springer-Verlag.



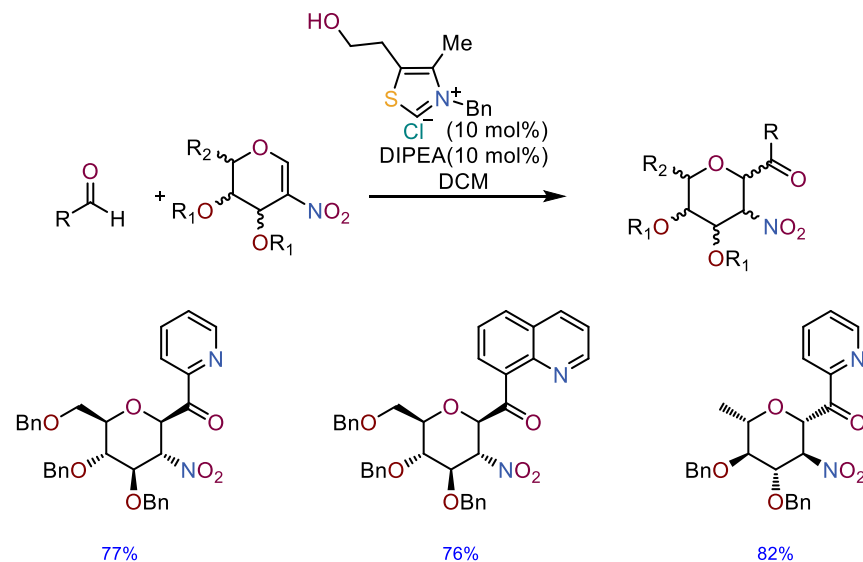
Enders, D. et al. *Chem. Commun.*, 2008, 3989–3991. <https://doi.org/10.1039/B809913H>

## Enantiopure $\alpha$ Amino Acid



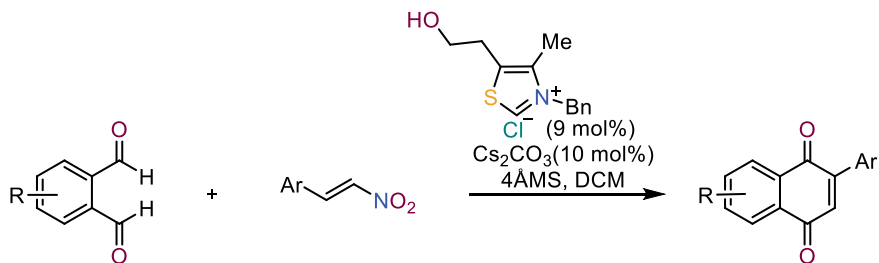
Glorius, F. et al. *Angew. Chem. Int. Ed.* 2011, 50, 1410–1414. <https://doi.org/10.1002/anie.201006548>

## C-Glycosylation



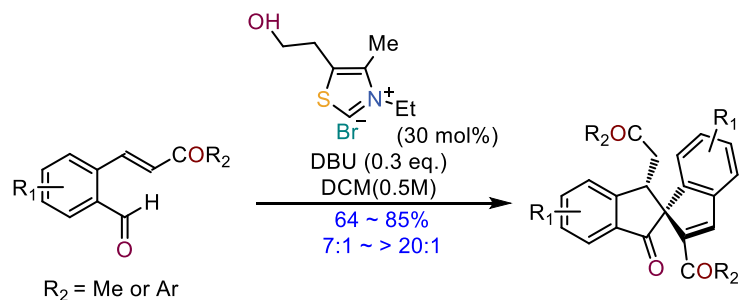
Liu, X. W. et al. *Org. Lett.* 2012, 14, 1, 174–177. <https://doi.org/10.1021/ol202959y>

## Tandem Reaction

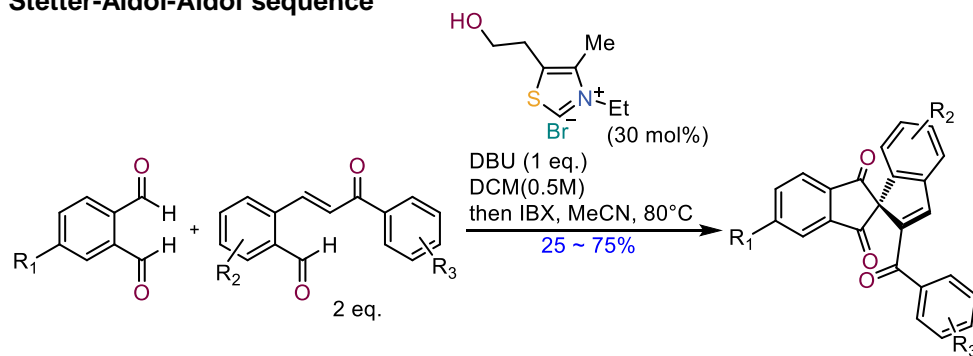


Maiti, D. K. et al. *J. Org. Chem.* 2019, 84, 1, 42–52. <https://doi.org/10.1021/acs.joc.8b01503>

## Stetter-Aldol-Michael sequence

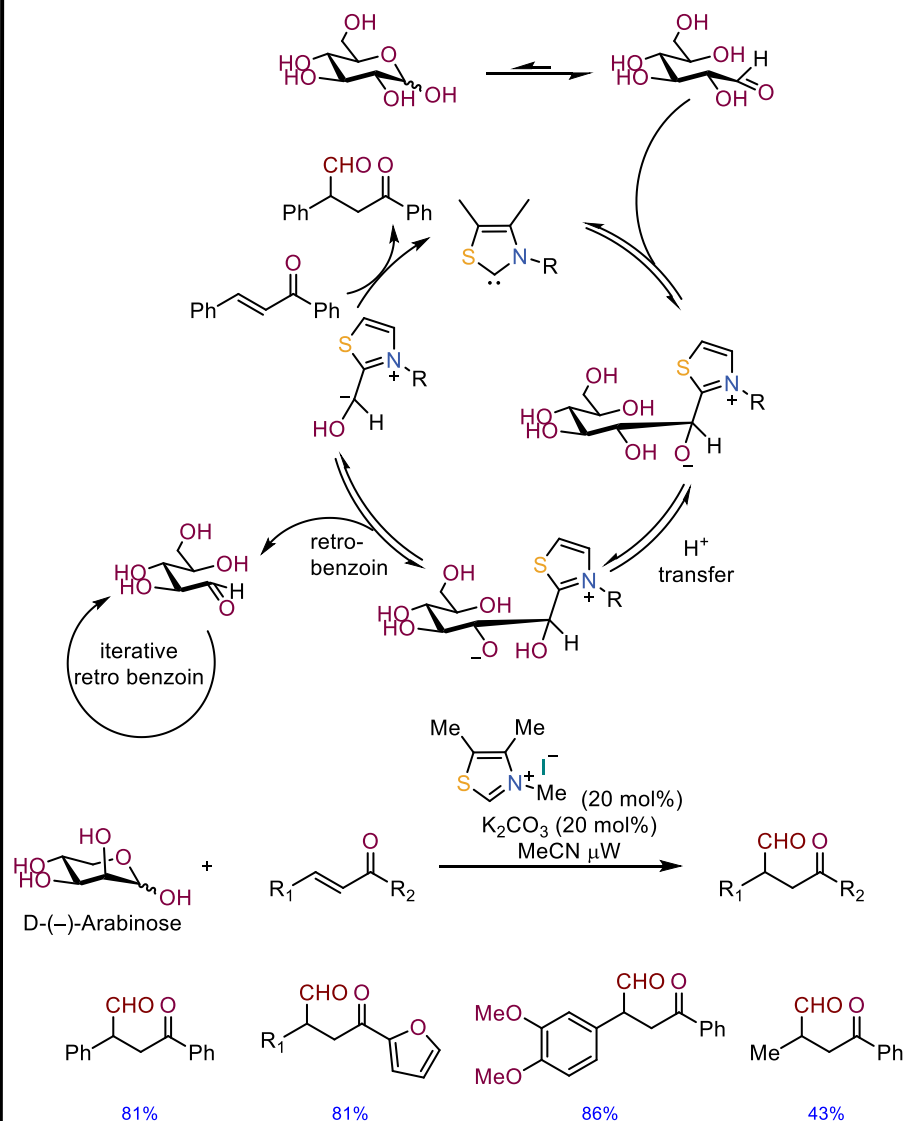


## Stetter-Aldol-Aldol sequence



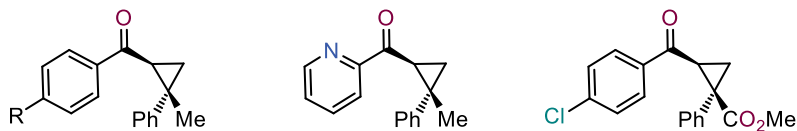
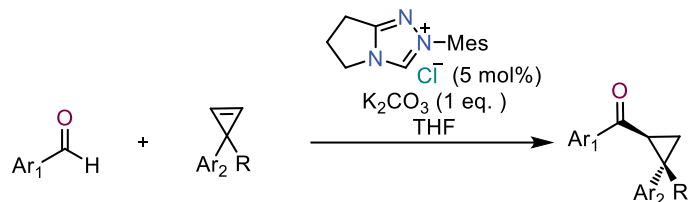
Gravel, M. Et al. *Org. Lett.* 2010, 12, 24, 5772–5775. <https://doi.org/10.1021/ol102685u>

## Monosaccharide as Formaldehyde Equivalent



Chi, Y. R. et al. *J. Am. Chem. Soc.* 2013, 135, 22, 8113–8116 <https://doi.org/10.1021/ja401511r>

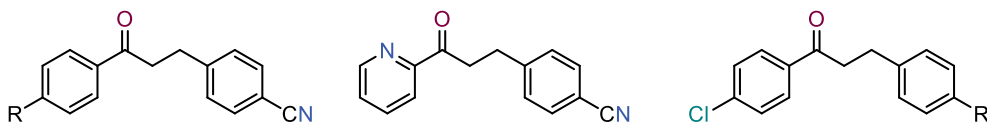
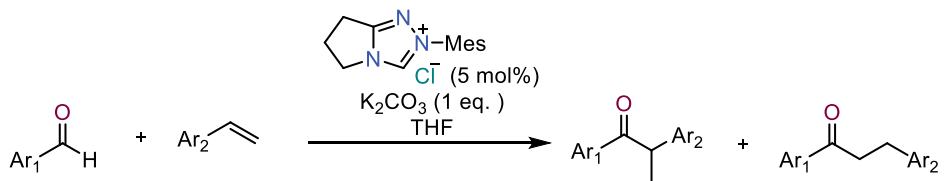
## Hydroacylation



$R = Br$  96%, dr>20:1  
 $R = CF_3$  78%, dr=11:1  
 $R = OMe$  44%, dr>20:1

Glorius, F. et al. *J. Am. Chem. Soc.* 2011, 133, 21, 8130–8133. <https://doi.org/10.1021/ja202594g>

## Hydroacylation

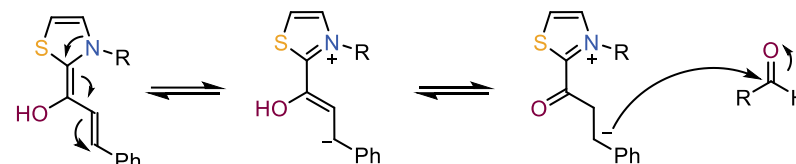


$R = Me$  47%  
 $R = Br$  73%, l:b=6:1  
 $R = Cl$  89%, l:b=5:1

$R = CO_2Me$  84%, l:b=3.9:1  
 $R = CF_3$  52%, l:b=1.1:1  
 $R = t-Bu$  39%, l:b=1:20  
 $R = O-t-Bu$  20%, l:b=1:20

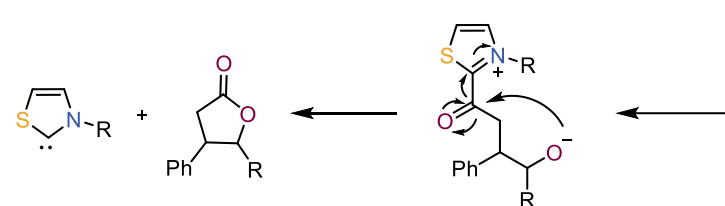
Glorius, F. et al. *Angew. Chem.* 2013, 125, 2645–2649. <https://doi.org/10.1002/ange.201209291>

## Homoenolate



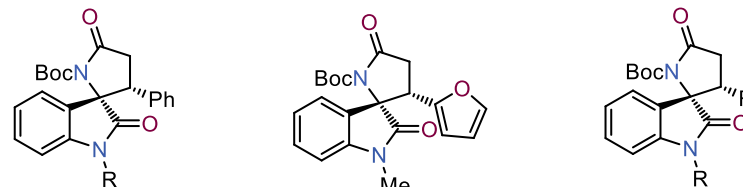
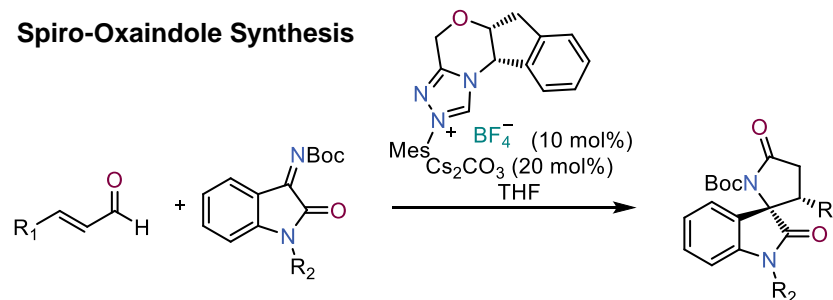
### Breslow intermediate

### Homoenolate



Glorius, F. et al. *Angew. Chem. Int. Ed.* 2004, 43, 6205–6208 <https://doi.org/10.1002/anie.200461572>  
 Bode, J. W. et al. *J. Am. Chem. Soc.* 2004, 126, 44, 14370–14371. <https://doi.org/10.1021/ja044714b>

## Spiro-Oxaindole Synthesis



$R = H$  64%, dr=17:1, 99% ee  
 $R = Me$  83%, dr>20:1, 99% ee  
 $R = OMe$  78%, dr=18:1, 99% ee

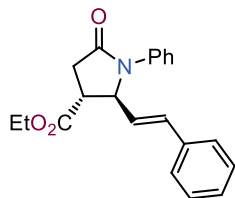
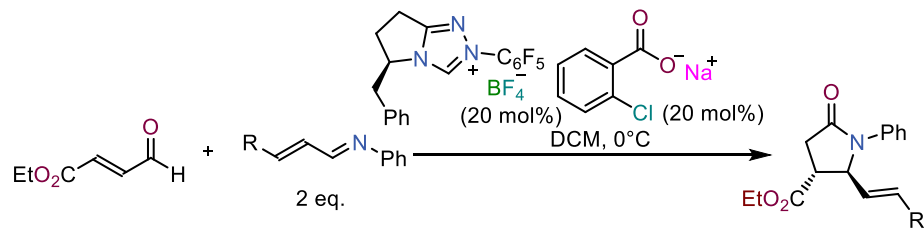
80%, dr>20:1  
 94% ee

$R = Me$  64%, dr=4:1, 94% ee  
 $R = n-Pr$  58%, dr=5:1, 99% ee

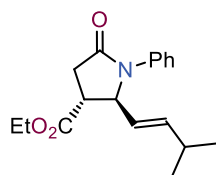
Chi, Y. R. et al. *Org. Lett.* 2012, 14, 21, 5412–5415. <https://doi.org/10.1021/ol302475g>

# Organocatalysis with NHCs

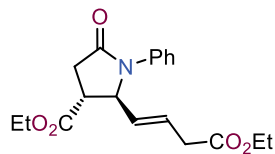
## Enantiopure $\gamma$ -lactam Synthesis [3+2] Annulation



92%, dr=14:1  
92% ee



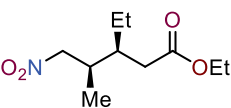
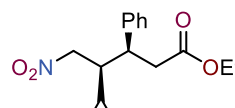
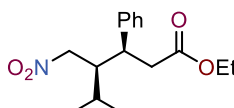
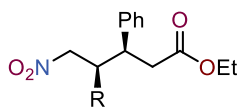
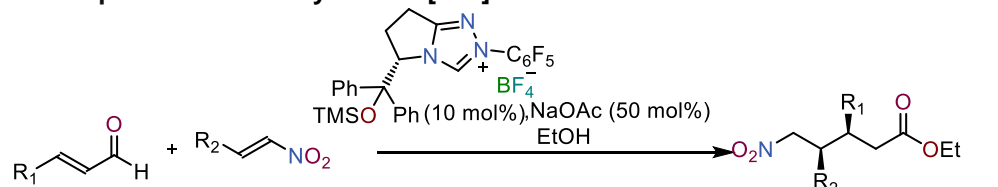
73%, dr>15:1  
88% ee



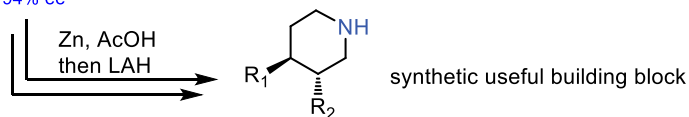
85%, dr=8:1  
92% ee

Rovis, T. et al. *J. Am. Chem. Soc.* 2011, 133, 32, 12466–12469. <https://doi.org/10.1021/ja205714g>

## Enantiopure - $\delta$ lactam Synthesis [3+3] Annulation

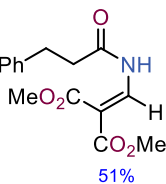
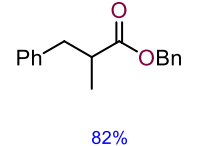
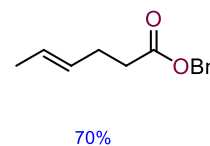
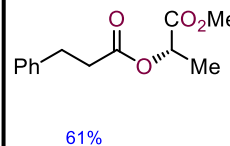
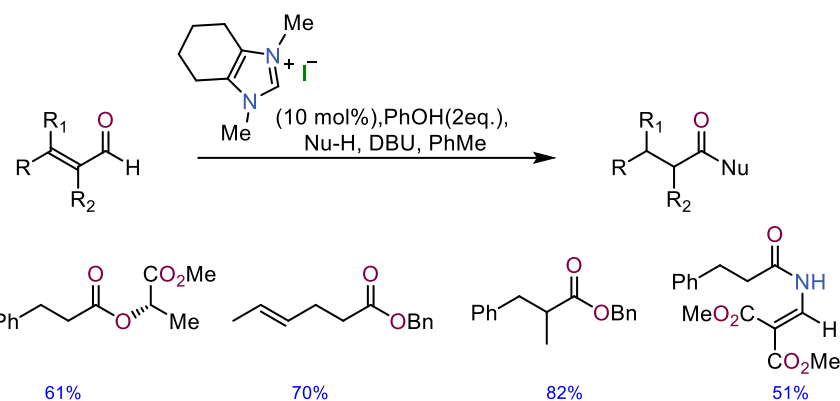


R=Me 60%, dr=17:1, 85% ee  
R=Et 70%, dr=17:1, 93% ee  
R=i-Pr 57%, dr=20:1, 94% ee



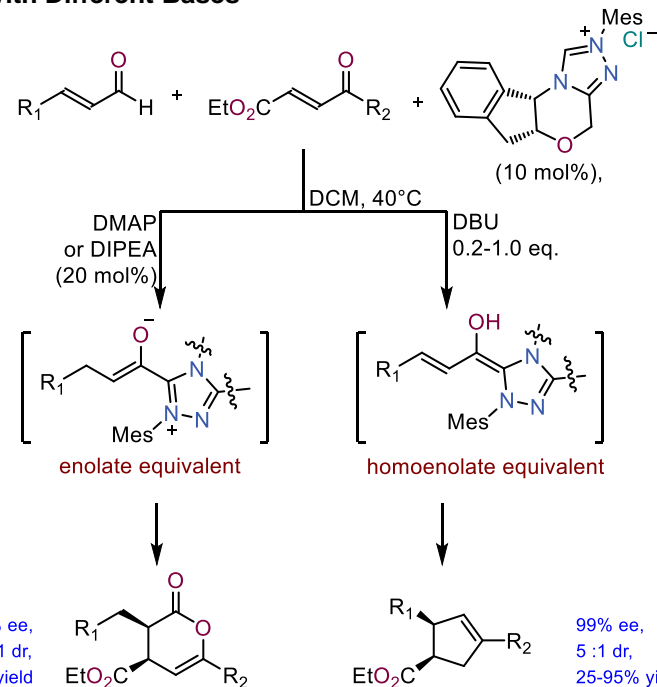
Rovis, T. et al. *J. Am. Chem. Soc.* 2013, 135, 23, 8504–8507. <https://doi.org/10.1021/ja403847e>

## $\beta$ Protonation

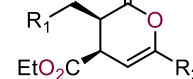


Scheidt, K. A. et al. *Org. Lett.* 2005, 7, 5, 905–908 <https://doi.org/10.1021/ol050100f>

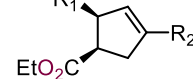
## Divergent with Different Bases



99% ee,  
> 20:1 dr,  
50–89% yield

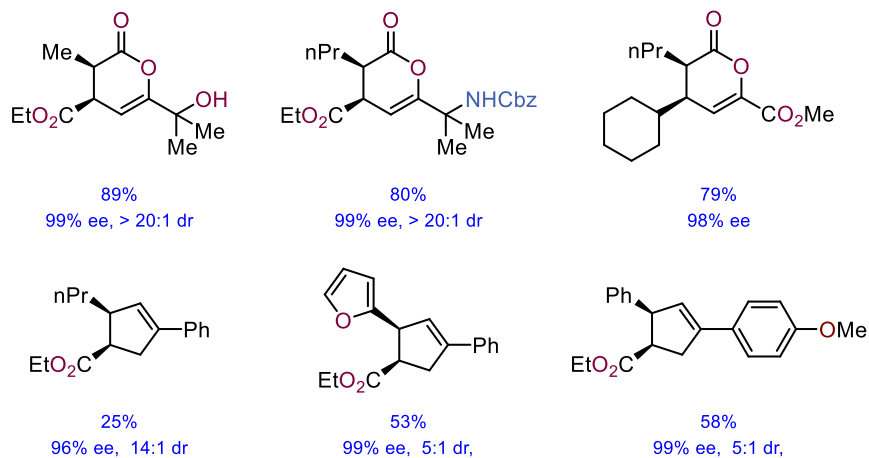


99% ee,  
5:1 dr,  
25–95% yield



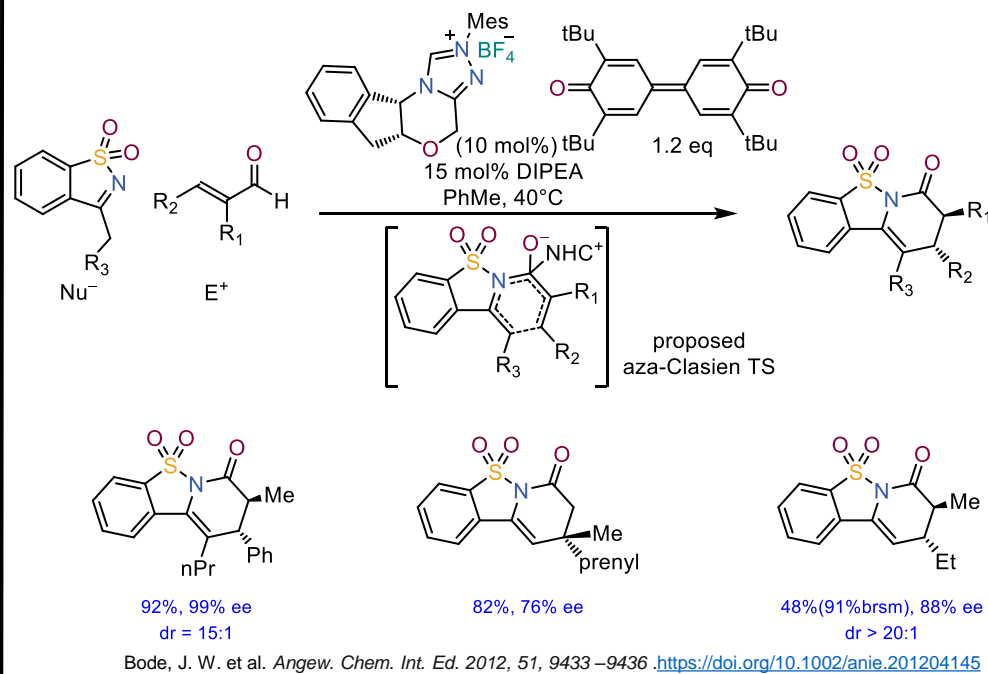
Bode, J. W. et al. *PNAS* 107 (48) 20661–20665. <https://doi.org/10.1073/pnas.1007469107>



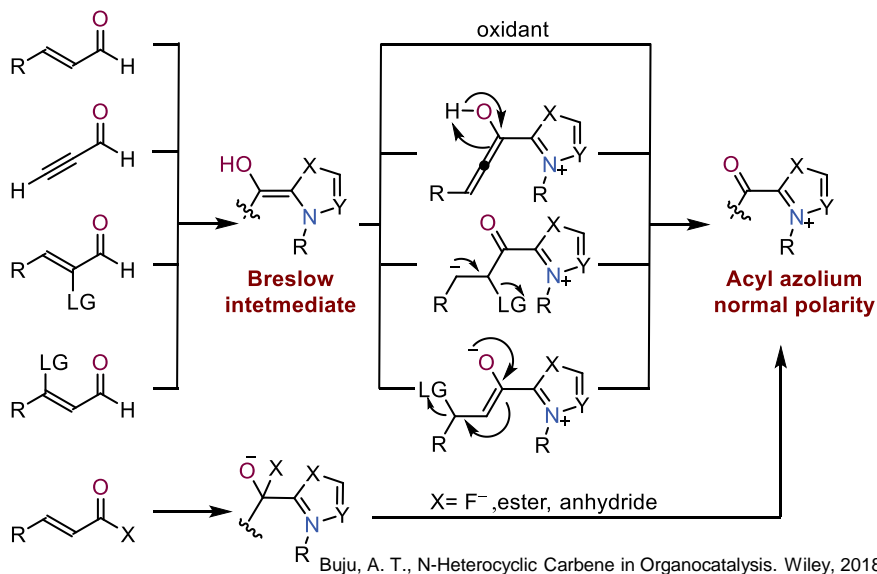


Bode, J. W. et al. *PNAS* 107 (48) 20661-20665. <https://doi.org/10.1073/pnas.1007469107>

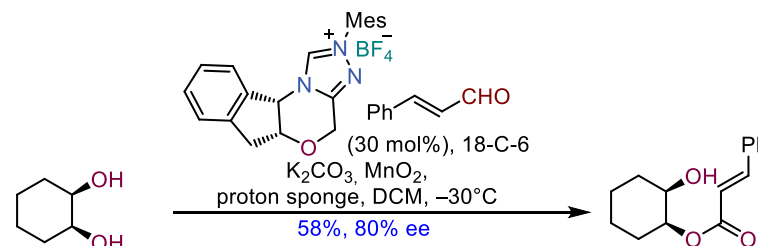
## Aza-Clasien



## Acyl Azolium

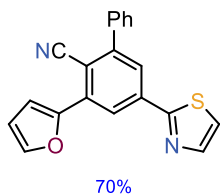
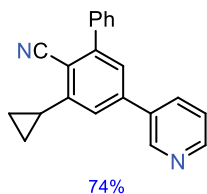
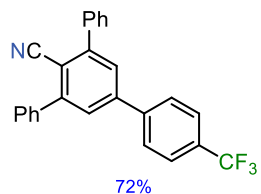
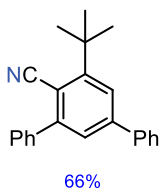
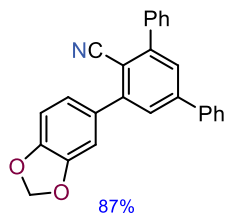
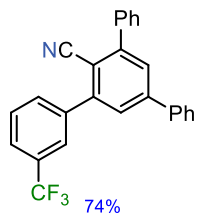
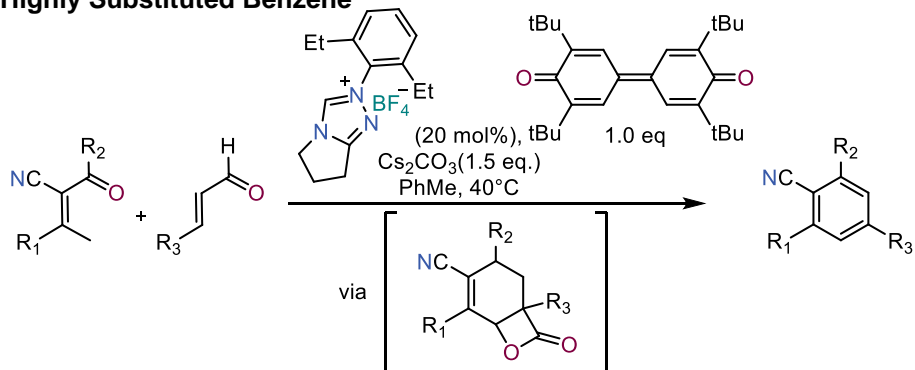


## Desymmetrization of 1,2-diol



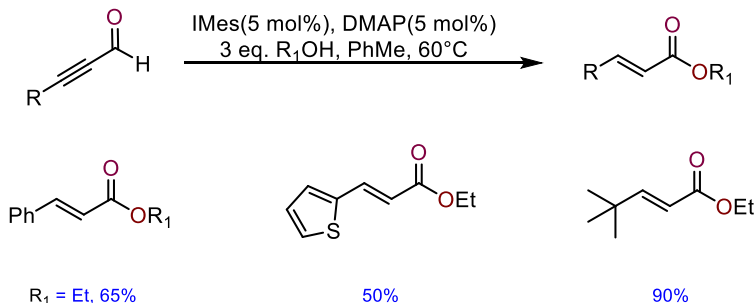
Scheidt, K. A. et al. *Org. Lett.* 2007, 9, 2, 371–374. <https://doi.org/10.1021/ol062940f>

## Highly Substituted Benzene



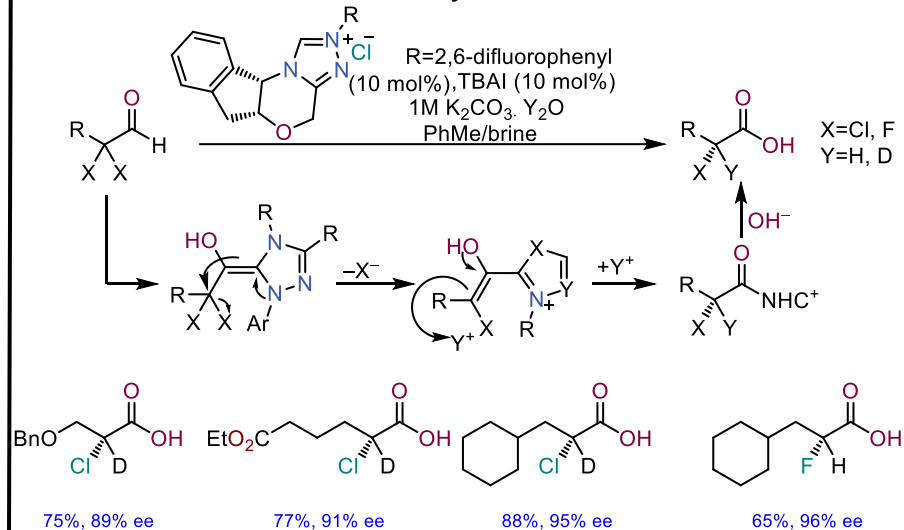
Wang, J. et al. *Org. Lett.* 2016, 18, 9, 2212–2215. <https://doi.org/10.1021/acs.orglett.6b00844>

## Redox Esterification



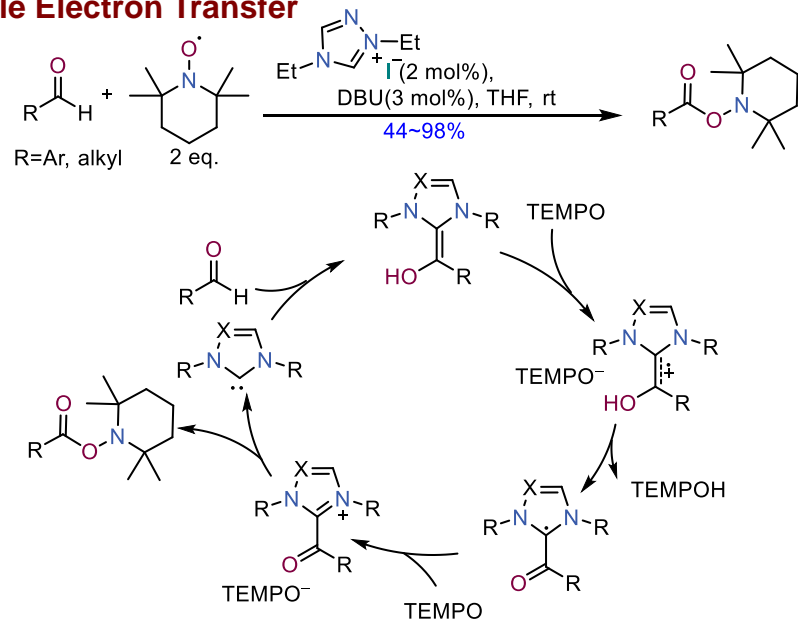
Zeitler, K. *Org. Lett.* 2006, 8, 4, 637–640. <https://doi.org/10.1021/ol052826h>

## Enantioselective $\alpha$ -chloro Carboxylic Acid



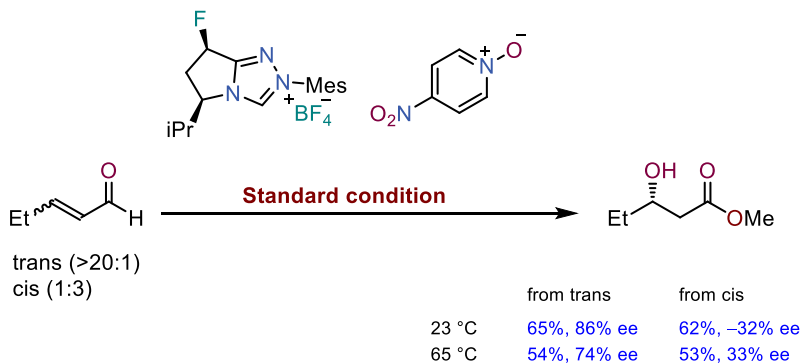
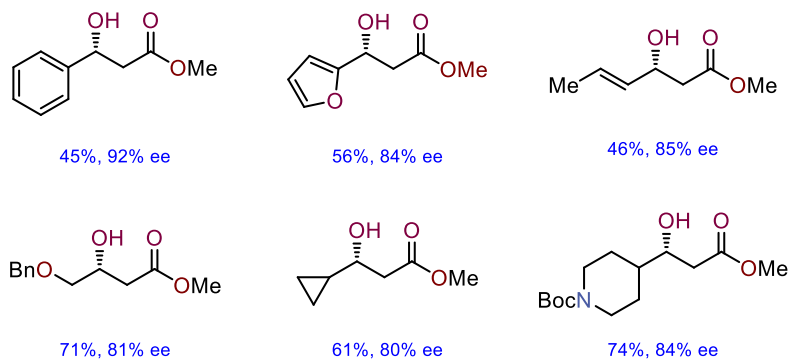
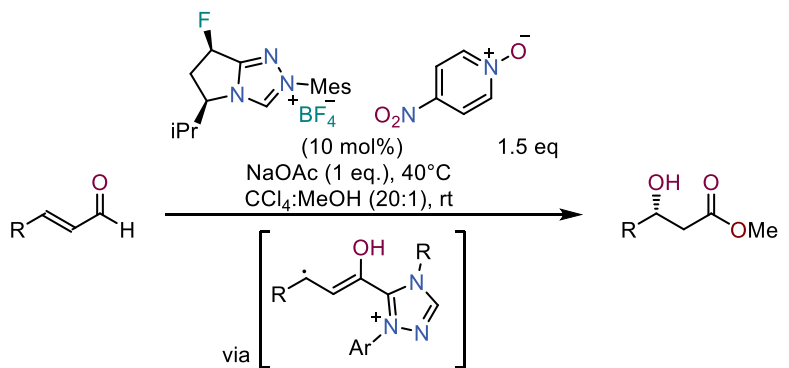
Rovis, T. et al. *J. Am. Chem. Soc.* 2010, 132, 9, 2860–2861. <https://doi.org/10.1021/ja910281s>

## Single Electron Transfer

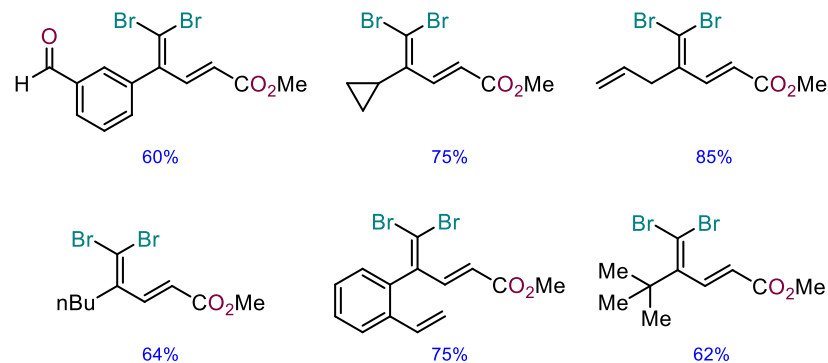
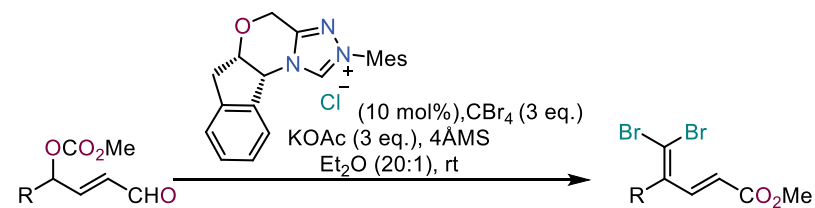


Studer, A. et al. *Angew. Chem. Int. Ed.* 2008, 47, 8727–8730. <https://doi.org/10.1002/anie.200802735>

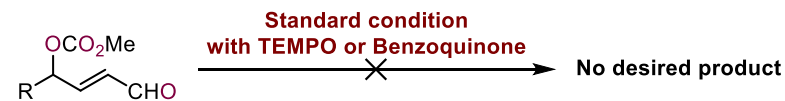
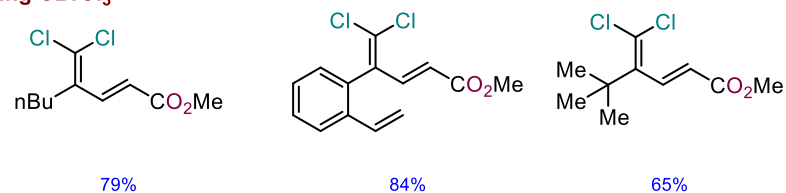
## Enantioselective $\beta$ -oxidation

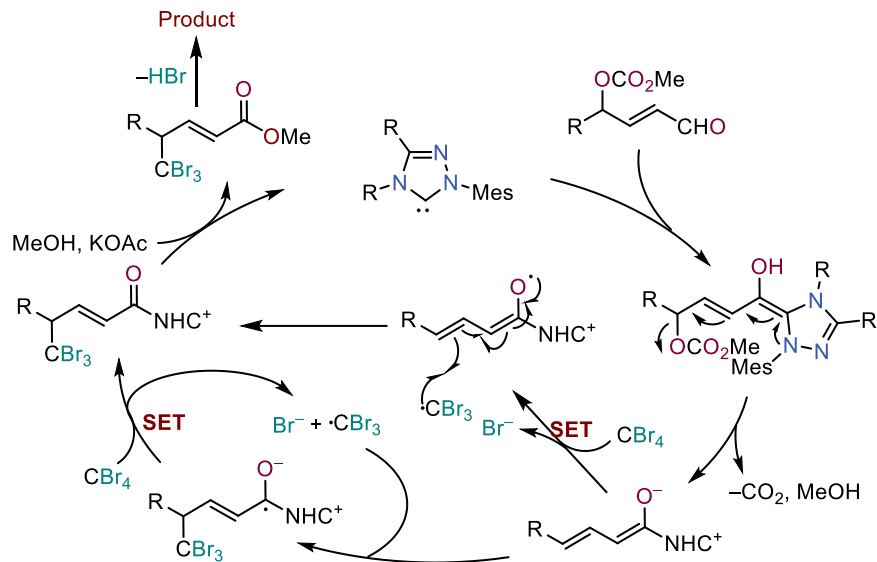


## Dibromomethylenation



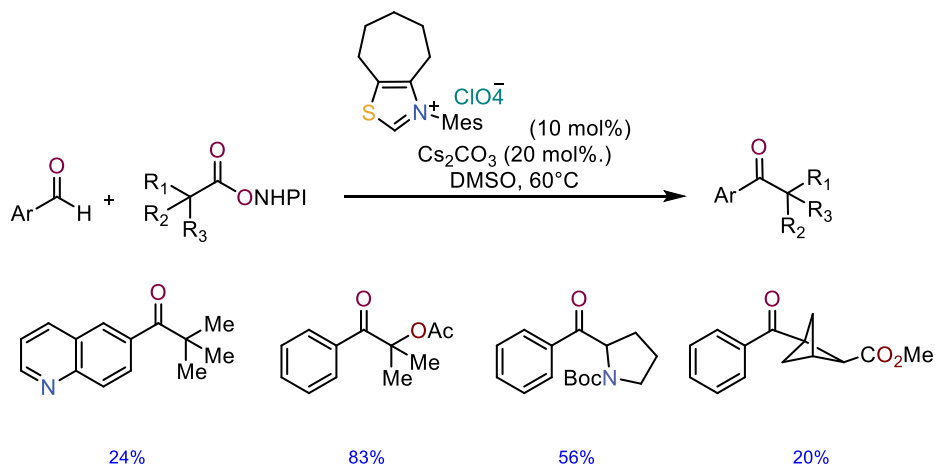
### If using $\text{CBrCl}_3$





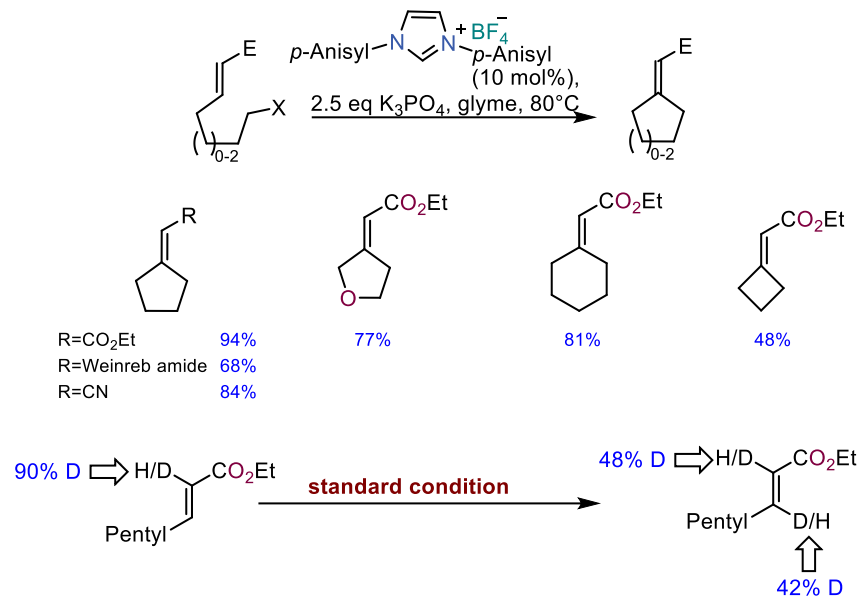
Sun, J. et al. *Angew. Chem. Int. Ed.* 2016, 55, 15783 – 15786 <https://doi.org/10.1002/anie.201608371>

## SET with Redox-Active Ester



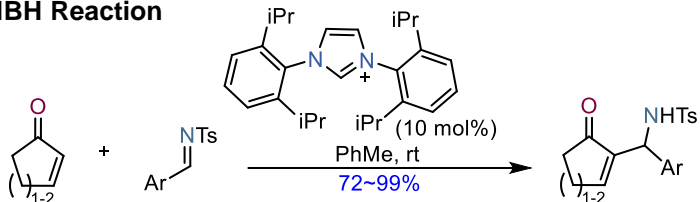
Ohmiya, H. et al. *J. Am. Chem. Soc.* 2019, 141, 9, 3854–3858. <https://doi.org/10.1021/jacs.9b00880>

## Umpolung MBH reaction



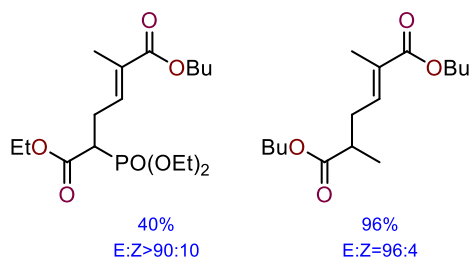
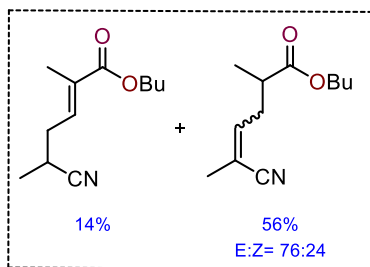
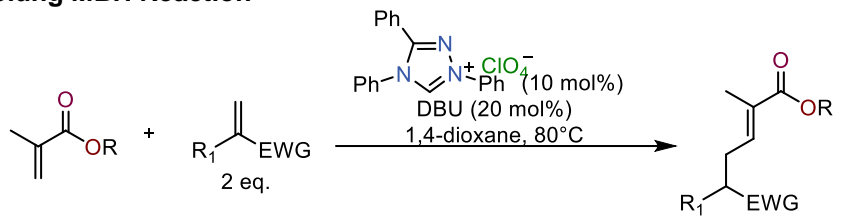
Fu, G. C. et al. *J. Am. Chem. Soc.* 2006, 128, 5, 1472–1473. <https://doi.org/10.1021/ja058222q>

## Aza-MBH Reaction



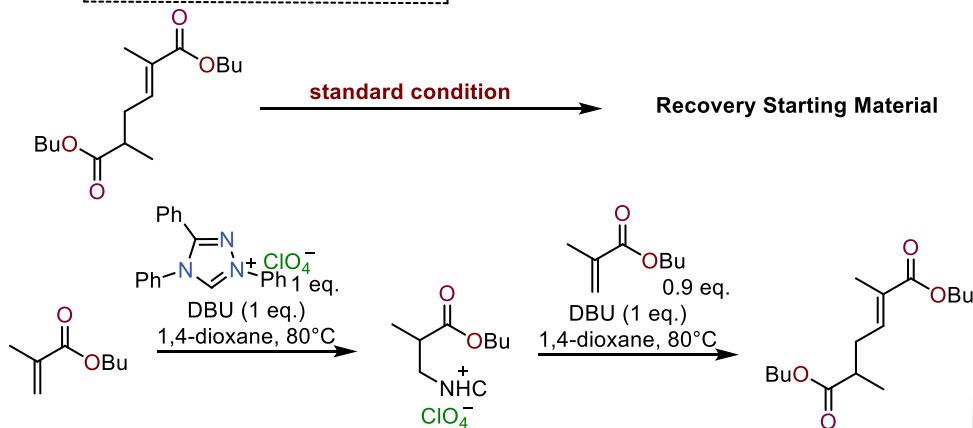
Ye, S. et al. *J. Org. Chem.* 2007, 72, 19, 7466–7468. <https://doi.org/10.1021/jo0712471>

## Umpolung MBH Reaction



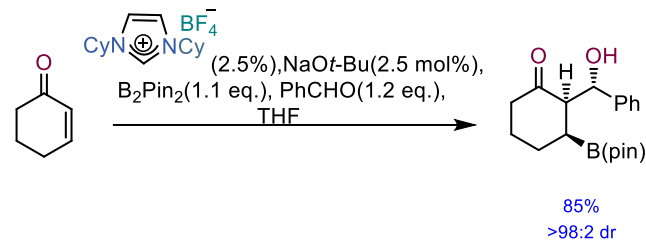
**standard condition**

**Recovery Starting Material**

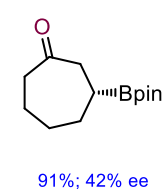
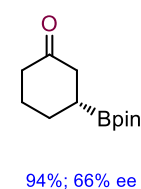
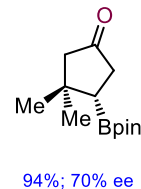
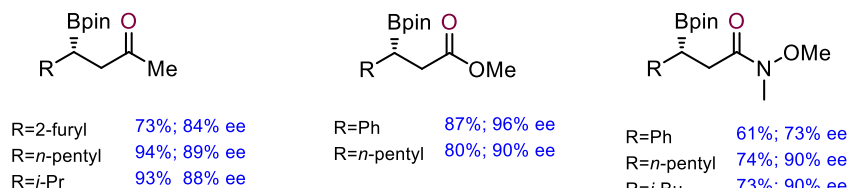
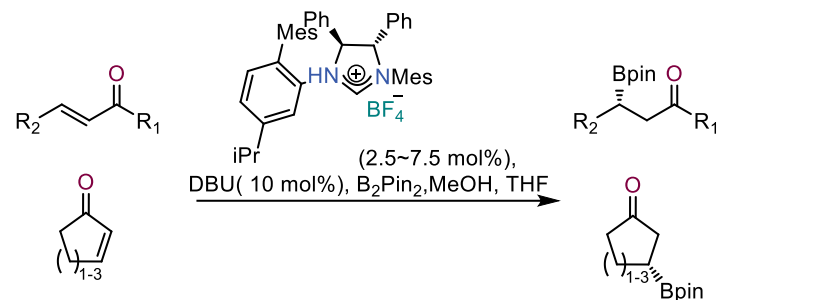


Glorius, F. et al. *Angew. Chem. Int. Ed.* 2011, 50, 8412–8415. <https://doi.org/10.1002/anie.201103555>

## Activation of Diboron



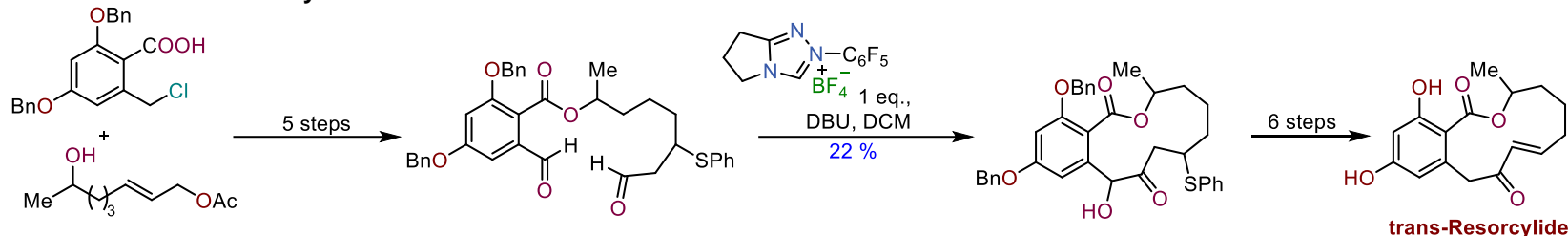
Hoveyda, A. H. et al. *J. Am. Chem. Soc.* 2009, 131, 21, 7253–7255. <https://doi.org/10.1021/ja902889s>



Hoveyda, A. H. et al. *J. Am. Chem. Soc.* 2015, 137, 33, 10585–10602. <https://doi.org/10.1021/jacs.5b06745>

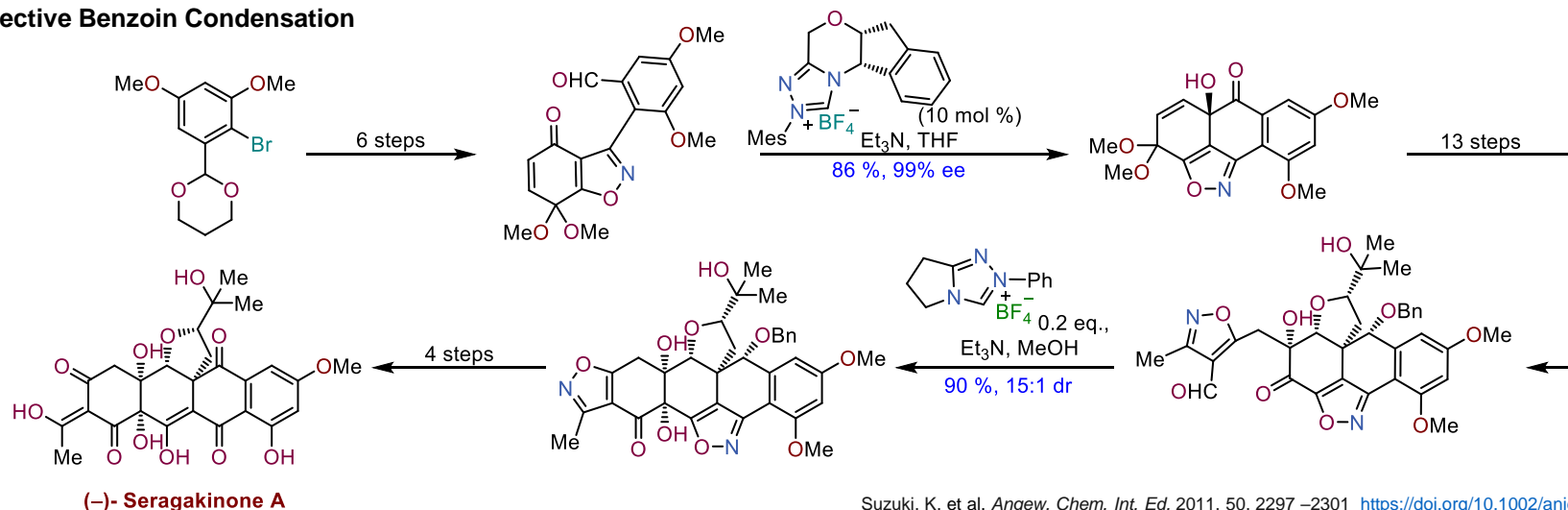
## Total synthesis examples with NHCs

### Benzoïn Condensation - Macrocyclization



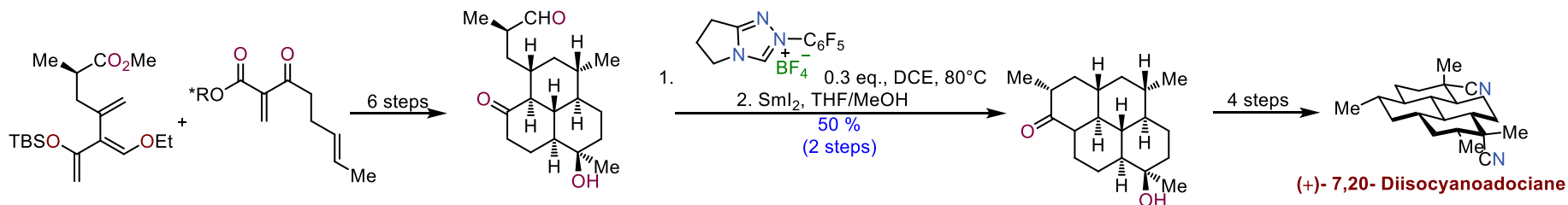
Miller, S. J. et al. *J. Org. Chem.* 2007, 72, 14, 5260–5269. <https://doi.org/10.1021/jo070676d>

### Enantioselective Benzoïn Condensation



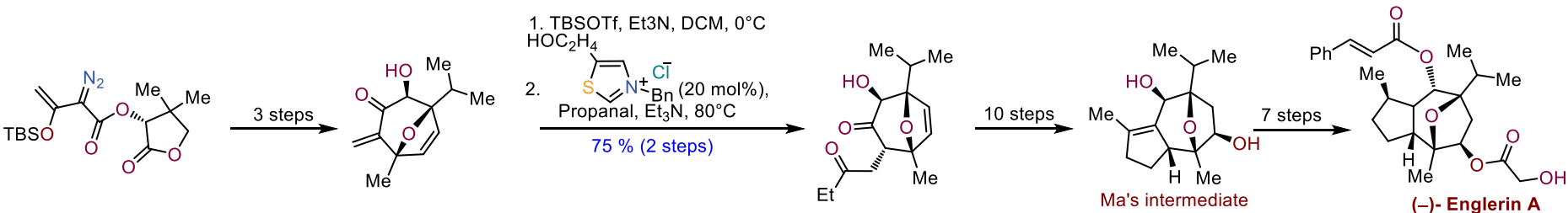
Suzuki, K. et al. *Angew. Chem. Int. Ed.* 2011, 50, 2297–2301 <https://doi.org/10.1002/anie.201006528>

### Benzoïn Condensation Keto-Aldehyde



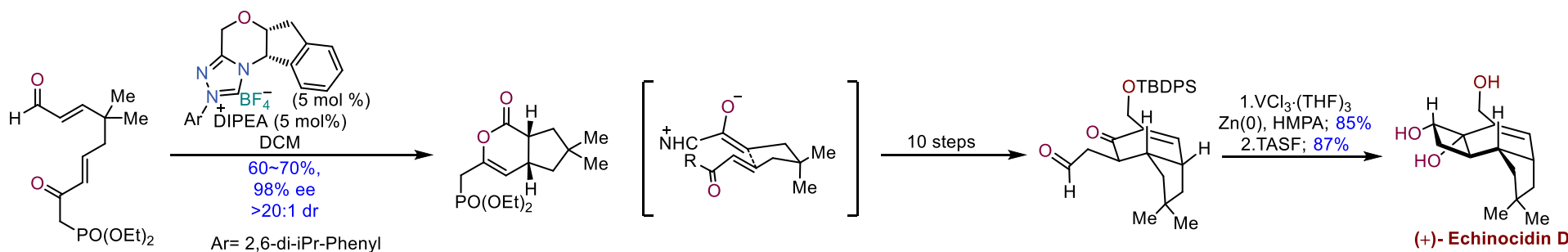
Shenvi, R. et al. *J. Am. Chem. Soc.* 2016, 138, 23, 7268–7271. <https://doi.org/10.1021/jacs.6b03899>

## Stetter reaction



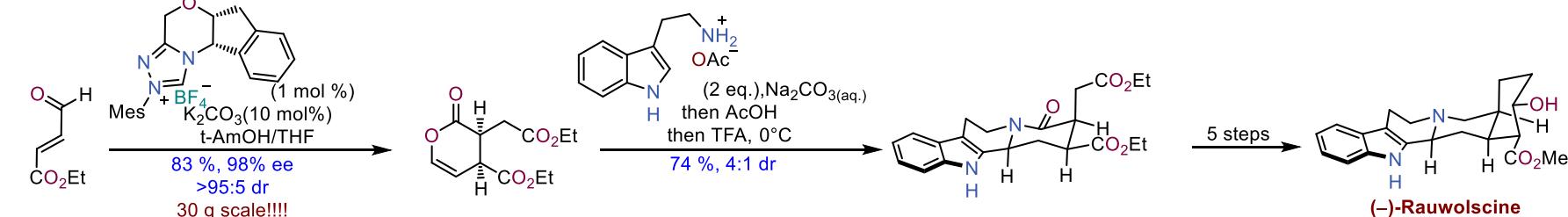
Theodorakis, E. A. et al. *Org. Lett.* 2010, 12, 16, 3708–3711. <https://doi.org/10.1021/ol1015652>  
Ma, D. et al. *Angew. Chem. Int. Ed.* 2010, 49, 3513–3516 <https://doi.org/10.1002/anie.201000888>

## Acyl Enolate



Scheidt, K. A. et al. *Angew. Chem. Int. Ed.* 2017, 56, 9864–9867. <https://doi.org/10.1002/anie.201705308>

## Acyl Enolate



Scheidt, K. A. et al. *J. Am. Chem. Soc.* 2020, 142, 5, 2187–2192. <https://doi.org/10.1021/jacs.9b12319>

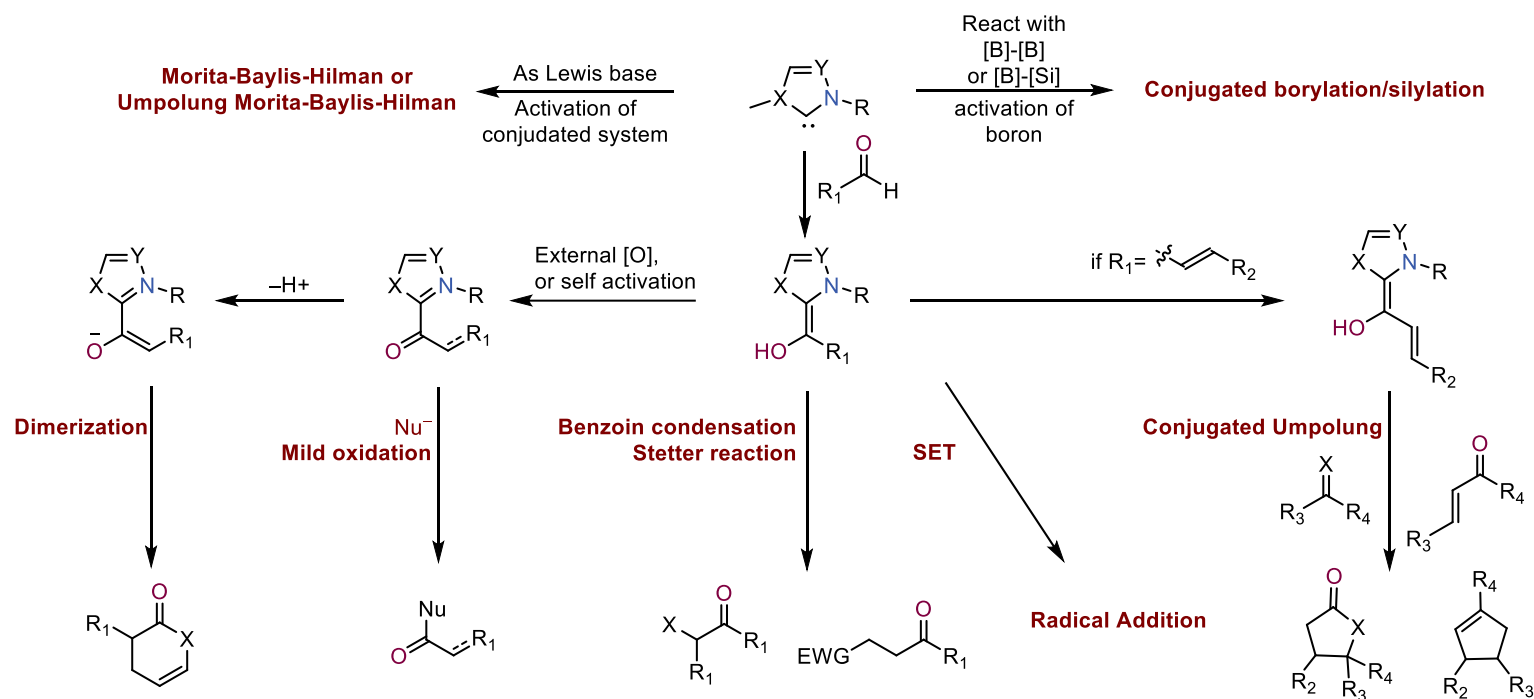
## Other Remarkable Total Synthesis with NHCs

- Kinamycin C, F, J (Nicolaou, 2007) (benzoin condensation with complex skeleton)
- Bryostatin 7 (Krische, 2011) (late-stage redox esterification)

Nicolaou, K. C. Et al. *J. Am. Chem. Soc.* 2007, 129, 34, 10356–10357. <https://doi.org/10.1021/ja074297d>  
Krische, M. J. et al. *J. Am. Chem. Soc.* 2011, 133, 35, 13876–13879. <https://doi.org/10.1021/ja205673e>

# Organocatalysis with NHCs

## Reactivity Map of NHC



Glorius, F. et al. *Nature*. 2014, 485–496. <https://doi.org/10.1038/nature13384>

### Advantage

- Versatile activation mode
- Fastly access complex skeleton
- Enable asymmetric synthesis

### Disadvantage

- Tailored substrate required
- No generation condition (reactivity depends on substrate, counter ion, stereoelectronic property of NHC)
- Activated substrates mostly (conjugated system)