

Background:

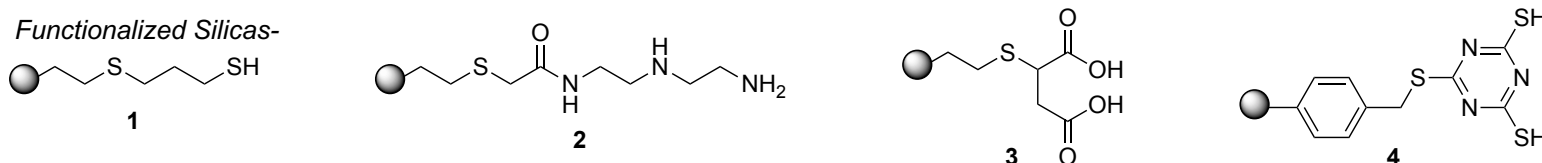
- Palladium-catalyzed cross-coupling reactions are used extensively
- In the pharmaceutical industry, strict regulatory limits have motivated the development of methods for metal scavenging
- Recovery of palladium can mitigate its high cost, limited supply, and environmental impact

Review: Economidou, M. *Org. Process Res. Dev.* **2023**, 27 (9), 1585–1615. <https://doi.org/10.1021/acs.oprd.3c00210>

Adsorption:

- Adsorption refers to the adhesion of particles onto a surface called the adsorbent
- Treatment of organic solutions with solid adsorbents (e.g. silica gels, resins, and activated carbon) allows filtration of the metal to give product-rich liquors
- Benefits: inexpensive, Limitations: formation of new impurities during treatment, leaching of impurities from the adsorbent, and yield losses

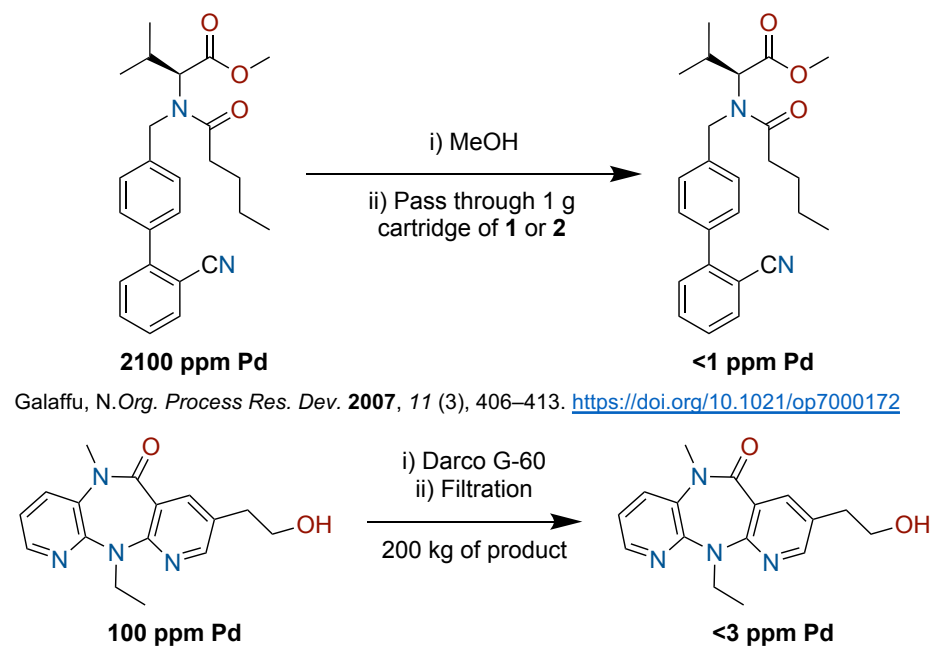
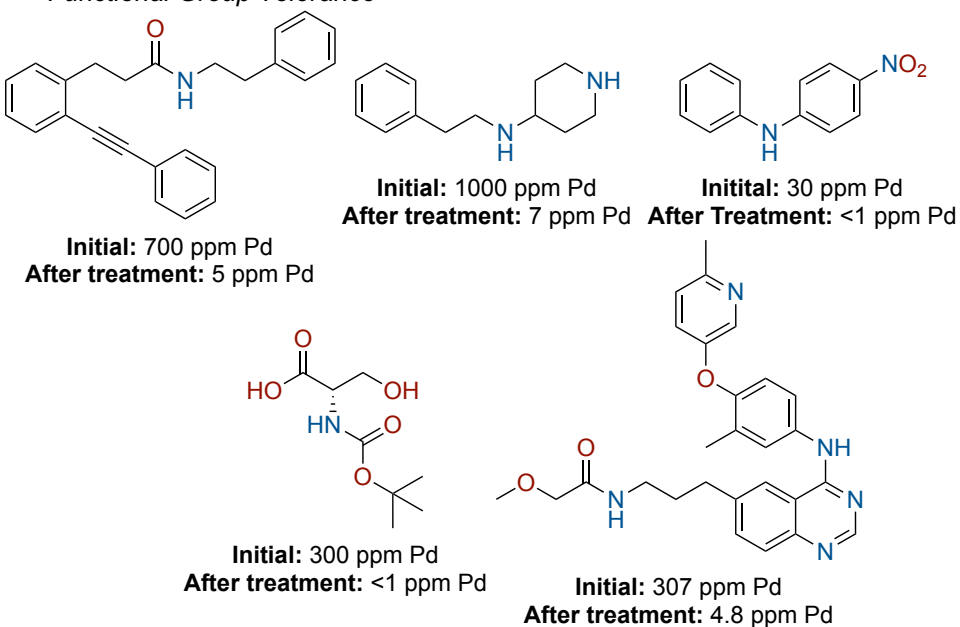
Functionalized Silicas-



Activated Carbons-

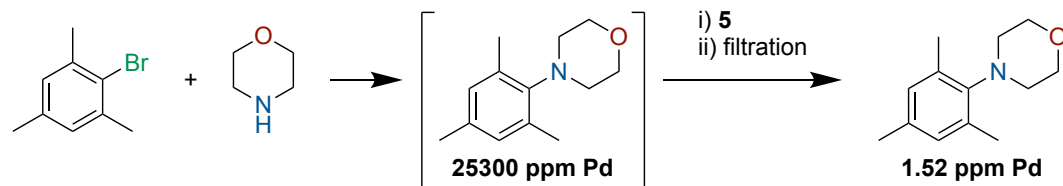
- Darco KB-B
- Ecosorb C-941

Functional Group Tolerance-

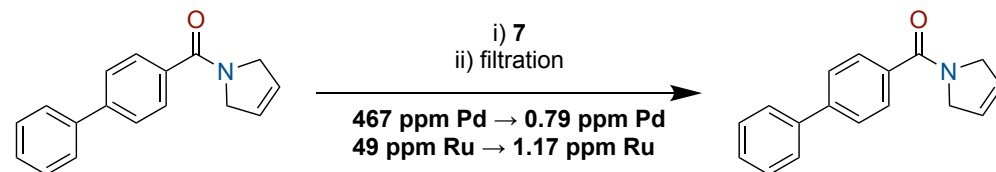
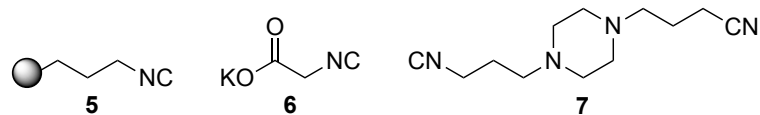


Isocyanides:

- First reported for the removal of ruthenium
- Operates by formation of an isocyanide insertion complex
- Homogenous scavenger **6** requires chromatography while silica-supported **5** does not



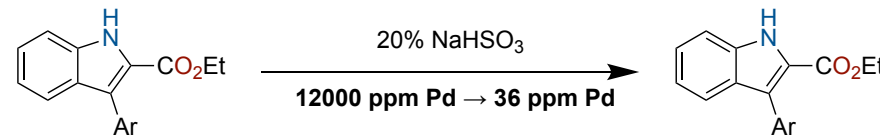
Common Scavengers-



French, J. M. *Adv. Synth. Catal.* **2015**, 357 (2–3), 361–365. <https://doi.org/10.1002/adsc.201400754>

Extraction/Precipitation:

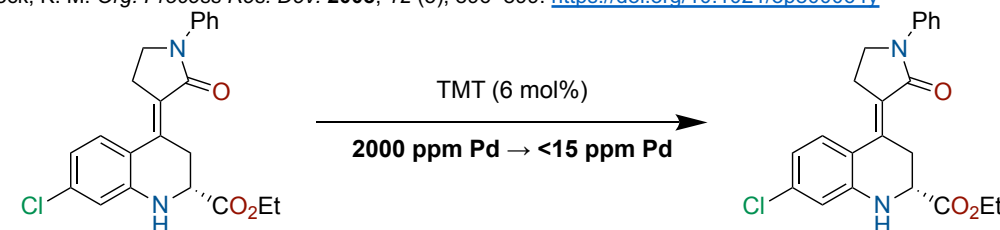
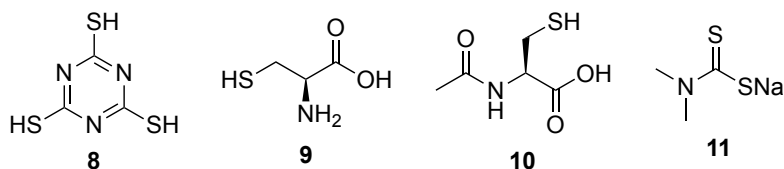
- Extraction involves chelation of palladium with a hydrophilic species to achieve transfer to the aqueous phase
- Precipitation involves formation of an insoluble metal complex allowing removal via filtration



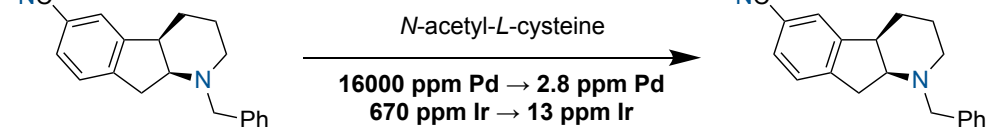
Bullock, K. M. *Org. Process Res. Dev.* **2008**, 12 (5), 896–899. <https://doi.org/10.1021/op800064y>

Common Agents-

- Sodium bisulfite
- Trimercaptotriazines, e.g. TMT (**8**)
- L-Cysteine (**9**) and N-acetyl cysteine (**10**)
- Dithiocarbamates (DTCs), e.g. sodium dimethyl dithiocarbamate (**11**)



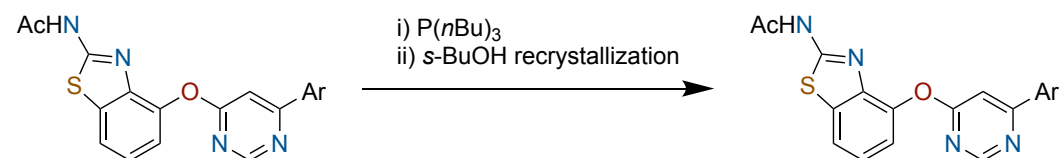
Banks, A. *Org. Process Res. Dev.* **2009**, 13 (6), 1130–1140. <https://pubs.acs.org/doi/10.1021/op9001824>



Qu, B. *Org. Process Res. Dev.* **2022**, 26 (3), 963–975. <https://doi.org/10.1021/acs.oprd.1c00290>

Crystallization:

- Crystallization is typically less efficient in the presence of metals
- Additives, including P(*n*Bu)₃, N-acetyl cysteine, L-cysteine, and xanthates, are often necessary



Larsen, R. D. *J. Org. Chem.* **1994**, 59 (21), 6391–6394. <https://doi.org/10.1021/jo00100a048>