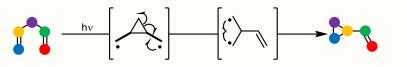


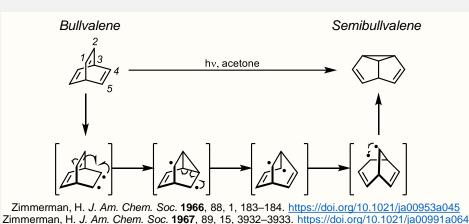


#### **General Scheme:**



Zimmerman, H. J. Am. Chem. Soc. 1969, 91, 7, 1718–1727. https://doi.org/10.1021/ja01035a021

## **Historical Development:**

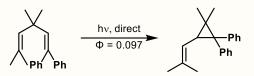


Mechanistically, the reaction was probed through the analysis of the hydrogen disposition in the product formed with a D-labeled bullvalene precursor.

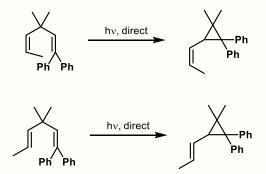
From this mechanism looking at the excited and ground states, the stabilization of the singlet or triplet excited states of the nonconjugated diene results when the orbitals of the C-2 and C-4 occurs. However, later studies would reveal that triplet sensitizers were unable to drive this reaction. Thus, it was determined only the singlet excited state manifold would efficiently drive this reaction forward. The excitation state of these reactions, however, is very substrate dependent.

The reaction can be summarized as the photolysis of molecules containing two  $\pi$  moleties bound to a single sp<sup>3</sup>-hybridized carbon atom. The reaction results in a  $\pi$ -substituted cyclopropane via a formal migration of one  $\pi$ -molety to the saturated carbon center.

## **Regioselectivity and Specificity Probes:**



The most stabilized radicals will undergo the final cyclopropanation, resulting in a single regioisomer.



The cis-isomer and trans isomers yielded exclusively their corresponding cis and trans counterparts.

Zimmerman, H. Chem. Rev. 1973, 73, 5, 531–551. https://doi.org/10.1021/cr60285a005

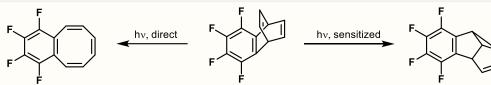
Zimmerman, H. J. Am. Chem. Soc. 1969, 91, 7, 1718–1727. https://doi.org/10.1021/ja01035a021



# Di-π-methane Rearrangement



## An Example of Irradiation Conditions Changing Products:

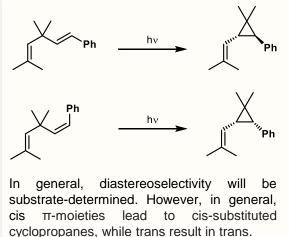


Depending on the methods of initiation, different products will form. The di-π-methane rearrangement above occurs under sensitized conditions only. Zimmerman, H. Chem. Rev. **1973**, 73, 5, 531–551. <u>https://doi.org/10.1021/cr60285a005</u>

#### **General Reaction Efficiency:**

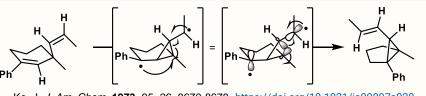
Di-π-methane structural feature	Singlet excited state	Triplet excited state
Presence of nonconstrained or free $\pi$ moiety (Acyclic and Monocyclic)	Efficient	Inefficient
Absence of nonconstrained or free $\pi$ moiety (Bicyclic)	Inefficient	Efficient

## **General Stereochemical Outcomes:**





Endo or exo products are generally not predictive but tend to be quite selective per substrate.



Ko, J. J. Am. Chem. 1973, 95, 26, 8670-8678. https://doi.org/10.1021/ja00807a028

Singlet states proceed via a concerted antidisrotatory mechanism while the triple state exhibits a two-step process which results in a formal anti-disrotatory mechanism.

Zimmerman, H. Chem. Rev. 1973, 73, 5, 531–551. https://doi.org/10.1021/cr60285a005

Schaffner, K. Helv. Chim. Acta 1980, 63, 2434–2439. DOI: 10.1002/hlca.19800630836

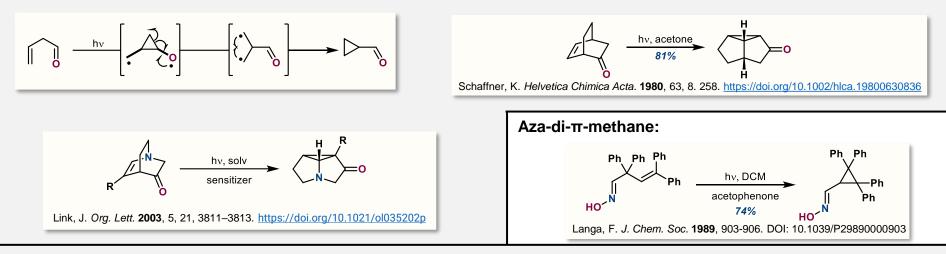
#### Jonathan Maturano



# Di-π-methane Rearrangement

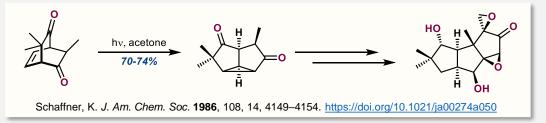


### Oxa-di-π-methane:



### **Examples in Total Synthesis:**

The Total Synthesis of (-)-Coriolin (Schaffner, 1986):



The Total Synthesis of Magellanine (Liao, 2002):

